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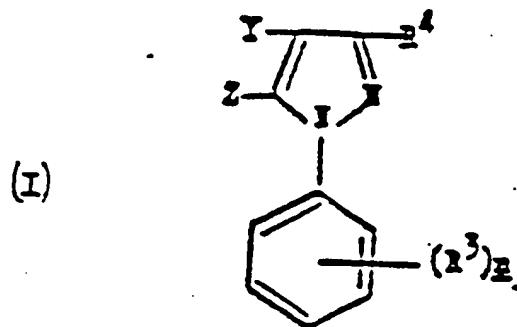
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(54) Title: PESTICIDAL METHOD USING N-PHENYL PYRAZOLES



(57) Abstract

A method for the control of arthropod, plant nematode or helminth pests using compounds of formula (I), wherein Y is halogen, cyano, nitro, RSO₂, RSO or RS in which R is alkyl, cycloalkyl or alkenyl, thiocyanato, sulphamoyl, carbamoyl, alkoxycarbonyl, alkanoyl or alkyl; Z represents hydrogen, an amino group -NR¹R², alkylsulphonyl amino, alkoxymethyleneamino, halogen, alkyl, carboxy, alkylthio, alkylsulphanyl or alkylsulphonyl, trialkylsilylmethyl, trialkylsilyl, alkano or nitro; R³ represents halogen, alkyl or alkoxy, halogen-substituted alkylthio or alkylsulphanyl, nitro, cyano, or alkylsulphonyl; and R⁴ represents halogen, cyano, nitro, alkyl or cycloalkyl, and n is 1 to 5, and salts thereof, provided that R⁴, Y and Z do not simultaneously represent one of nitro, cyano, halogen and unsubstituted alkyl; pesticidal compositions, novel compounds and processes for their preparation are described.

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DESCRIPTION

"PESTICIDAL METHOD USING N-PHENYL PYRAZOLES"

This invention relates to the use of N-phenylpyrazole derivatives against arthropod, plant nemacode and helminth pests, to compositions containing them and to novel N-phenylpyrazole derivatives.

In J. Heter. Chem., 12 (1975), 1199-1205, P.L. Southwick and B. Dhawan have described experiments for the preparation of 4,6-diamino-pyrazolo[3,4-d]pyrimidines in the expectation that such pyrimidine derivatives would have useful pharmacological properties. They employed as starting materials 5-amino-4-cyanopyrazoles carrying on the 1-position a hydrogen atom, a methyl group, a hydroxyethyl group or a phenyl group substituted by one or more chlorine atoms and/or methyl groups, and on the 3-position a hydrogen atom, a methyl group or a phenyl or benzyl group. This publication contains no suggestion that compounds of general formula I possess or would be expected to possess activity against arthropods, helminths or plant nematodes.

Apparently these pyrazole compounds did not lead (according to the authors of the article) to useful therapeutic (viz. antimalarial) 4,6-diaminopyrazolo[3,4-d]pyrimidines.

US Patent No. 3760084 describes certain 5-amino-1-phenyl-pyrazoles as being useful for ameliorating inflammation in warm-blooded animals: the compounds carry on the 3-position hydrogen or a lower alkyl group and on the 4-position a carbamoyl or cyano group.

US Patent No. 3869274 describes certain 4-nitropyrazoles as being useful for the induction of abscission of fruit from fruit-bearing plants.

US Patent No. 4066776 describes a very extensive group of 1,4-disubstituted-3-nitropyrazoles as having antimicrobial, parasiticidal and herbicidal properties: the great biological activity of the compounds is stated to be limited to the 3-nitropyrazoles disclosed, the characterising feature of the compounds being the 3-nitropyrazole nucleus.

Japanese Patent Publication No. 12644/64 describes a process for the preparation of 4-thiocyanatopyrazole derivatives which are stated to be useful as germicides.

Japanese Patent Publication No. 49-117502 describes certain pyrazole sulphonamides having anti-thrombogenic properties.

None of the foregoing publications describes or suggests that compounds of general formula I possess or would be expected to possess the activity against arthropods, helminths or plant nematodes which has been discovered by the Applicants.

It has now unexpectedly been found after extensive research and experimentation that the N-phenylpyrazole derivatives of the general formula I depicted hereinafter wherein Y represents a halogen, i.e. 5 fluorine, chlorine, bromine or iodine, ~~accm~~, the cyano or nitro group or a group RSO_2 , RSO or RS in which R represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which may be unsubstituted or substituted by one or more halogen 10 atoms, a cycloalkyl group containing from 3 to 5 carbon atoms, a straight- or branched-chain alkenyl group containing from 2 to 6 carbon atoms, the thiocyanato group, the sulphamoyl group which may be unsubstituted or substituted by one or two straight- 15 or branched-chain alkyl groups which may be the same or different and contain from 1 to 6 atoms, the carbamoyl group which may be unsubstituted or substituted by one or two straight- or branched-chain alkyl groups which may be the same or different and contain from 1 to 6 carbon atoms, a straight- or 20 branched-chain alkoxy carbonyl group containing from 2 to 7 carbon atoms, a straight- or branched-chain alkanoyl group containing from 2 to 7 carbon atoms, or a straight- or branched-chain alkyl group containing 25 from 1 to 6 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms.

Z represents one hydrogen atom, or the amino group -NR¹R² wherein R¹ and R², which may be the same or different, each represents a hydrogen atom or a straight- or branched-chain alkyl group (containing from 1 to 6 carbon atoms, and which may be unsubstituted or substituted by straight- or branched-chain alkoxy carbonyl of 2 to 5 carbon atoms), cycloalkyl group containing from 3 to 6 carbon atoms, formyl group, straight- or branched-chain alkanoyl group (which contain from 2 to 7 carbon atoms or together form a 5 or 6 membered cyclic imide with the nitrogen atom to which they are attached and themselves may be unsubstituted or substituted by one or more halogen atoms) or cycloalkyl carbonyl group (which contain from 4 to 7 carbon atoms) or straight- or branched-chain alkoxy carbonyl groups (which contain from 2 to 7 carbon atoms and themselves are unsubstituted or substituted by one or more halogen atoms), or Z represents a straight- or branched-chain alkylsulphenylamino group containing from 1 to 4 carbon atoms, a straight- or branched-chain alkoxymethyleneamino group containing from 2 to 5 carbon atoms which may be unsubstituted or substituted on methylene by a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms, or represents a halogen, i.e. fluorine, chlorine, bromine or iodine, atom, a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms, the carboxy

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group, or a straight- or branched-chain alkylthio, alkylsulphinyl or alkylsulphocyl group containing from 1 to 6 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms, or

5 represents a straight- or branched-chain trialkylsilylmethyl group containing from 1 to 6 carbon atoms in each alkyl group which may be the same or different, a trialkylsilyl group containing from 1 to 6 carbon atoms in each alkyl group which may be

10 the same or different or the cyano or nitro group.

R^3 represents a halogen, i.e. fluorine, chlorine, bromine or iodine atom, a straight- or branched-chain alkyl or alkoxy group containing from 1 to 4 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms, (e.g. a trifluoromethyl or

15 trifluoromethoxy group), a straight- or branched-chain alkylthio or alkylsulphinyl group containing from 1 to 4 carbon atoms which is substituted by one or more halogen atoms (e.g. a trifluoromethylthio or

20 trifluoromethylsulphinyl group), the nitro or cyano group or a straight- or branched-chain alkylsulphonyl group containing from 1 to 4 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms (e.g. the trifluoromethylsulphonyl group), and

25 R^4 represents a halogen, i.e. fluorine, chlorine, bromine or iodine atom, a cyano or nitro group or a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms which may be unsubstituted or

substituted by one or more halogen atoms, or a cycloalkyl group containing from 3 to 6 carbon atoms, and n represents an integer from 1 to 5 inclusive, and, when Z represents a carboxy group, salts thereof with pesticidally-acceptable bases provided that R^4 , Y and Z do not simultaneously represent three groups of the same genus selected from the genera (i) nitro, (ii) cyano, (iii) halogen and (iv) unsubstituted alkyl, have valuable activity against arthropod, plant nematode and helminth pests, more particularly by ingestion of the compound(s) of general formula I by the arthropods. When n represents an integer from 2 to 5 inclusive, atoms and groups represented by R^3 may be the same or different.

By the term 'salts with pesticidally acceptable bases' is meant salts the cations of which are known and accepted in the art for the formation of salts of pesticidally active acids for agricultural or horticultural use. When intended for application to vertebrates to combat infection or infestation by arthropods or helminths, the salts with bases used will be non-toxic. By the term 'non-toxic' is meant salts with bases the cations of which are innocuous to the vertebrates at the doses administered and which do not vitiate the beneficial effects produced by the anion.

Preferably, the salts are water-soluble.

Suitable salts with bases include alkali metal (e.g. sodium and potassium), alkaline earth metal (e.g. calcium and magnesium), ammonium and amine (e.g. diethanolamine, triethanolamine, ocyamine, morpholine and diocetylamine) salts. It is to be understood that where reference is made in the present specification to the compounds of general formula I such reference is intended to include also the salts with pesticidally acceptable bases of compounds of general formula I where appropriate.

Preferred compounds of general formula I are those with phenyl substitution which is 2,4,6-trichloro, 2,3,5,6-tetrachloro, 2-chloro-4-trifluoromethyl, 15 2,3,5,6-tetrafluoro-4-trifluoromethyl, 2,6-dichloro-4-trifluoromethylthio, 2-chloro-3,5,6-trifluoro-4-trifluoromethyl, 2,6-dichloro-3,5-difluoro-4-trifluoromethyl, 2,6-dichloro-4-nitro, 2,6-dichloro-4-trifluoromethyl-20 sulphanyl, 2,6-dichloro-4-methanesulphonyl and 2,6-dichloro-4-trifluoromethanesulphonyl.

Compounds of general formula I wherein (R^3) represents 2,6-dichloro-4-trifluoromethyl or 2,6-dichloro-4-trifluoromethoxy substitution of the phenyl group are especially preferred.

Preferred compounds are those where

5 (a) Y and R⁴ each represent a cyano group and
Z represents the hydrogen atom, the amino group
-NR¹R² or an alkylsulphenylamino group, an
alkoxymethyleneamino group which may be
unsubstituted or substituted on methylene by an
alkyl group, a halogen atom, an alkyl group, the
carboxy group, an alkylthio, alkylsulphanyl or
alkylsulphonyl group which is optionally halogen
10 substituted, a trialkylsilylmethyl group, a
trialkylsilyl group or the nitro group;

15 (b) Y represents an alkylsulphonyl group which is
optionally halogen substituted, a
cycloalkylsulphonyl group or an alkenylsulphonyl
group, Z represents the hydrogen atom, the amino
group -NR¹R² or an alkylsulphenylamino group,
an alkoxymethyleneamino group which is
unsubstituted or substituted on methylene by an
alkyl group, a halogen atom, an alkyl group, the
carboxy group, an alkylthio, alkylsulphanyl or
alkylsulphonyl group which is optionally halogen
20 substituted, a trialkylsilylmethyl group, a
trialkylsilyl group or the cyano or nitro group
and R⁴ represents a halogen atom or the cyano
or nitro group;

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(c) R^4 represents the nitro group, Y represents the cyano or nitro group, a carbamoyl group or an alkoxy carbonyl group and
Z represents the hydrogen atom, a halogen atom, an alkyl group, the carboxy group, an alkylthio, alkylsulphinyl or alkylsulphonyl group which is optionally halogen substituted, a trialkylsilylmethyl group, a trialkylsilyl group or the nitro group;

5 (d) R^4 represents a halogen atom, Y represents the cyano or nitro group, a carbamoyl group or an alkoxy carbonyl group and
Z represents the hydrogen atom, the amino group $-NR^1R^2$ or an alkylsulphenylamino group, an alkoxy methylene amino group which is unsubstituted or substituted on methylene by an alkyl group, a halogen atom, an alkyl group, the carboxy group, an alkylthio, alkylsulphinyl or alkylsulphonyl group which is optionally halogen substituted, a trialkylsilylmethyl group, a trialkylsilyl group or the nitro group; and

10 (e) R^4 represents an alkyl group which is unsubstituted or substituted by one or more halogen atoms or a cycloalkyl group, Y represents a halogen atom, the cyano or nitro group, a group RSO_2 , RSO or RS , the thiocyanato group, a sulphamoyl group, a carbamoyl group, an

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alkoxycarbonyl group, an alkanoyl group or an alkyl group which is unsubstituted or substituted by one or more halogen atoms,

5 Z represents the hydrogen atom, the amino group $-\text{NR}^1\text{R}^2$ or an alkylsulphenylamino group, an alkoxymethylenesamino group which is unsubstituted or substituted on methylene by an alkyl group, a halogen atom, an alkyl group, the carboxy group, an alkylthio, alkylsulphanyl or alkylsulphonyl 10 group which is optionally halogen substituted, a trialkylsilylmethyl group, a trialkylsilyl group or the cyano or nitro group.

15 It will be appreciated that the groups listed above are as hereinbefore defined earlier in the specification.

Compounds of general formula I wherein R^4 represents a trifluoromethyl or methyl group are also preferred.

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Compounds of general formula I which are of particular interest against arthropods are:

- 1 5-Amino-3,4-dicyano-1-(2,4,6-trichlorophenyl)pyrazole
- 5 2 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3,4-dicyanopyrazole
- 3 3 5-Amino-3,4-dicyano-1-(2,3,5,6-tetrachloro-phenyl)pyrazole
- 4 4 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoro-methylphenyl)-3-methylpyrazole
- 10 5 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoro-methylphenyl)-3-trifluoromethylpyrazole
- 6 6 5-Amino-3-chloro-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole
- 15 7 5-Amino-3-bromo-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole
- 8 8 5-Amino-3-iodo-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole
- 9 9 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-5-ethanesulphenylamino-pyrazole
- 20 10 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-5-methoxymethyleamino-pyrazole

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11 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-5-propoxymethyleneamino-pyrazole

12 5-Acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole

13 5-Dichloroacetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole

14 5-Cyclopropylcarbonamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole

10 15 5-Pantanamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole

16 5-Propionamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole

17 5-Amino-1-(2-chloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole

15 18 5-Amino-3,4-dicyano-1-(2,3,5,6-tetrafluoro-4-trifluoromethylphenyl)pyrazole

19 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-pentafluoroethylpyrazole

20 20 5-Amino-3-chlorodifluoromethyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyanopyrazole

21 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-3-difluoromethylpyrazole

25 22 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole

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23 5-Amino-4-carbamoyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole

24 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methoxycarbonyl-3-trifluoromethyl-pyrazole

25 5-Acetamido-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole

10 26 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-(2,2-dimethylpropionamido)-pyrazole

27 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-ethoxymethyleneamino-3-trifluoromethylpyrazole

15 28 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-dimethylamino-3-trifluoromethyl-pyrazole

29 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-ethoxycarbonylmethylamino-3-trifluoromethylpyrazole

20 30 4-Cyano-5-methylamino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole

25 31 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-(2,2-dimethylpropionamido)-3-trifluoromethylpyrazole

32 5-Amino-4-bromo-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole

33 5-Bromo-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole

34 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-fluoromethyl-pyrazole

35 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole

36 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethoxyphenyl)-3-trifluoromethyl-pyrazole

37 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-3-trifluoromethylpyrazole

38 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(cyclopropanecarbonyl)amino-3-trifluoromethylpyrazole

39 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-cyclopropanecarbonamido-3-trifluoromethylpyrazole

40 5-Amino-4-chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole

41 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-oxycarbonylamino-3-trifluoromethylpyrazole

42 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole

5 43 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-iodo-3-trifluoromethylpyrazole

44 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-methyl-3-trifluoromethylpyrazole

10 45 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-(N,N-dimethylsulphamoyl)-3-trifluoromethylpyrazole

46 5-Amino-4-cyano-3-cyclopropyl-1-(2,6-dichloro-4-trifluoromethyl-phenyl)pyrazole

15 47 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-heptafluoropropylpyrazole

48 5-Amino-3,4-dicyano-1-(2,6-dichloro-4-trifluoromethylthiophenyl)pyrazole

20 49 5-Amino-1-(2-chloro-3,5,6-trifluoro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole,

50 5-Amino-1-(2,6-dichloro-3,5-difluoro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole,

25 51 5-Amino-1-(2,6-dichloro-4-trifluoromethoxy-phenyl)-3,4-dicyanopyrazole,

52 5-Amino-4-cyano-3-ethyl-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole

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53 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methanesulphonyl-3-methylpyrazole

54 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-ethoxycarbonylpyrazole

55 5-Amino-1-(2,6-dichloro-4-trifluoromethoxy-phenyl)-4-methanesulphonyl-3-methylpyrazole

56 5-Amino-1-(2-chloro-3,5,6-trifluoro-4-trifluoromethylphenyl)-4-cyano-3-trifluoromethylpyrazole

10 57 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylthiophenyl)-3-trifluoromethylpyrazole

58 5-Amino-3-chlorofluoromethyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole

59 5-Amino-4-cyano-1-(2,6-dichloro-3,5-difluoro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole

15 60 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-(1-ethoxyethylideneamino)-3-methylpyrazole

61 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-5-succinimidopyrazole

62 5-Acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole

20 63 5-Acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-methanesulphonylpyrazole

25

64 5-Amino-1-(2,6-dichloro-4-nitrophenyl)-3,4-dicyanopyrazole

65 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-methylaminopyrazole

5 66 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-ethylaminopyrazole

67 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-(N-methyl-N-echoxycarbonylamino)-3-trifluoromethylpyrazole

10 68 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-(N-acetyl-N-trimethylacetylaminio)-3-trifluoromethylpyrazole

69 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-(N-propionyl-N-trimethylacetylaminio)-3-trifluoromethylpyrazole

15 70 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethyl-5-trimethylacetylaminopyrazole

71 1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-echoxycarbonylamino-4-nitro-3-trifluoromethylpyrazole

20 72 3-Chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-5-trimethylacetylaminopyrazole

73 3-Chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-5-bis(echoxy-carbonyl)aminopyrazole

74 3-Chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-5-ethoxycarbonylaminopyrazole

5 75 4-Cyano-5-diacetylamino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole

76 5-(N-Acetyl-N-ethoxycarbonylamino)-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole

10 77 1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-3,4-dicyanopyrazole

78 1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-4-methanesulphonyl-3-trifluoromethylpyrazole

15 79 1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-ethoxycarbonylaminoc-4-methanesulphonyl-3-trifluoromethylpyrazole

80 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-ethoxycarbonylaminopyrazole

20 81 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ido-3-trifluoromethylpyrazole

82 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ido-3-methylpyrazole

25 83 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-nitropyrazole

84 5-Acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole

85 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole

86 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3-methyl-4-methanesulphonylpyrazole

87 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-fluoropyrazole

10 88 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole

89 5-Chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-3-trifluoromethylpyrazole

15 90 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-(N-ethylsulphamoyl)-3-trifluoromethylpyrazole

91 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-(N-methylsulphamoyl)-3-trifluoromethylpyrazole

20 92 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-cyano-3-nitropyrazole

93 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-nitropyrazole

25 94 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-3-fluoropyrazole

95 5-Amino-3-chloro-1-(2,6-dichloro-4-trifluoromethoxyphenyl)-4-cyanopyrazole
96 5-Amino-3-chloro-4-cyano-1-(2,6-dichloro-3,5-difluoro-4-trifluoromethylphenyl)pyrazole
5 97 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-5-trimethylsilylpyrazole
98 5-tert-Butyldimethylsilyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole
10 99 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-methylthio-3-trifluoromethylpyrazole
100 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-5-trifluoromethylthiopyrazole
15 101 5-Carboxy-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole
102 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethyl-5-trimethylsilylpyrazole
20 103 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-5-trimethylsilylmethylpyrazole
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114 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-methoxycarbonylamino-3-trifluoromethylpyrazole

115 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4,5-dicyano-3-trifluoromethylpyrazole

116 5-Amino-3-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonylpyrazole

117 4-Acetyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole

118 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methylsulphiny-3-trifluoromethylpyrazole

119 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylsulphiny-3-trifluoromethylpyrazole

120 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylsulphiny-3-methylpyrazole

121 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylsulphinyphenyl)-3-trifluoromethylpyrazole

122 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-methylsulphiny-3-trifluoromethylpyrazole

123 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylsulphonyl-3-trifluoromethylpyrazole

114 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-ethylsulphonyl-3-methylpyrazole
115 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-propanesulphonyl-pyrazole
5
116 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-trichloromethanesulphonyl-3-methylpyrazole
117 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-ethylthio-3-methylpyrazole
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118 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-methylthiopyrazole
119 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4- α -propylthio-3-methylpyrazole
15
120 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-ethylthio-3-trifluoromethyl-pyrazole
121 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methylthio-3-trifluoromethyl-pyrazole
20
122 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-thiocyanato-3-trifluoromethyl-pyrazole
123 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-thiocyanatopyrazole
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124 5-Amino-4-cyano-1-(2,6-dichloro-4-methane-sulphonylphenyl)-3-trifluoromethyl-pyrazole

5 125 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-trichloromethylthio-pyrazole

126 4-Cyano-1-(2,6-dichloro-4-trifluoro-methanesulphonylphenyl)-5-nitro-3-trifluoromethylpyrazole

10 127 1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-difluoromethyl-3-trifluoromethyl-pyrazole

128 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methyl-3-trifluoromethyl-pyrazole

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The numbers 1 to 128 are assigned to the above compounds for identification and reference hereinafter.

Especially preferred compounds of general formula I are numbered:-

20 2; 22; 37; 53; 71; 106 and 118.

According to a feature of the present invention, there is provided a method for the control of arthropod, plant nematode or helminth pests at a locus which comprises the treatment of the locus (e.g. by application or administration) with an effective amount of a compound of general formula I, or a pesticidally acceptable salt thereof, wherein the various symbols are as hereinbefore defined. The compounds of general formula I may, in particular, be used in the fields of veterinary medicine and livestock husbandry and in the maintenance of public health against arthropods or helminths which are parasitic internally or externally upon vertebrates, particularly warm-blooded vertebrates, for example man and domestic animals, e.g. cattle, sheep, goats, equines, swine, poultry, dogs, cats and fishes, for example Acarina, including ticks (e.g. Ixodes spp., Boophilus spp. e.g. Boophilus microplus, Amblyomma spp., Hyalomma spp., Rhipicephalus spp. e.g. 15 Rhipicephalus appendiculatus, Haemaphysalis spp., Dermacentor spp., Ornithodoros spp. (e.g. Ornithodoros moubata) and mites (e.g. Damalinia spp., Dermanyssus gallinae, Sarcopotes spp. e.g. Sarcopotes scabiei, Psoroptes spp., Chorioptes spp., Demodex spp., 20 Eutrombicula spp.); Diptera (e.g. Aedes spp., Anopheles spp., Musca spp., Hypoderma spp., Gasterophilus spp., Simulium spp.);

Hemiptera (e.g. Triatoma spp.); Pachitacera (e.g. Damalinia spp., Lincognathus spp.); Siphonaptera (e.g. Ctenocephalides spp.); Dicyoptera (e.g. Periplaneta spp., Blatella spp.); Hymenoptera (e.g. Monomorium pharaonis); for example against infections of the gastro-intestinal tract caused by parasitic nematode worms, for example members of the family Trichostrongylidae, Nippostrongylus brasiliensis, Trichinella spiralis, Haemonchus contortus, Trichostrongylus colubriformis, Nematodirus battus, Ostertagia circumcincta, Trichostrongylus axei, Cooperia spp. and Hymenolepis nana; in the protection of stored products, for example cereals, including grain and flour, groundnuts, animal feedstuffs, timber and household goods, e.g. carpets and textiles, against attack by arthropods, more especially beetles, including weevils, moths and mites, for example Ephestia spp. (flour moths), Anthrenus spp. (carpet beetles), Tribolium spp. (flour beetles), Sitophilus spp. (grain weevils) and Acarus spp. (mites), in the control of cockroaches, ants and similar arthropod pests in infested domestic and industrial premises and in the control of mosquito larvae in waterways, wells, reservoirs or other running or standing water; in agriculture, against adults, larvae and eggs of Lepidoptera (butterflies and moths), e.g. Heliothis spp. such as Heliothis virescens (tobacco budworm),

Heliothis armigera and Heliothis zea, Scodotera spp.
such as S. exemota, S. littoralis (Egyptian cotton
worm), S. eridania (southern army worm), Manestra
configurata (bertha army worm); Earias spp. e.g.
E. insulana (Egyptian bellworm), Pectinophora spp. e.g.
Pectinophora gossypiella (pink bollworm), Ostrinia
spp. such as O. nubilalis (European cornborer),
Trichocelusia ni (cabbage looper), Pieris spp. (cabbage
worms), Lachryma spp. (army worms), Agrotis and
Amathes spp. (cutworms), Wiseana spp. (porina moth),
Chilo spp. (rice stem borer), Tryporyza spp. and
Diatraea spp. (sugar cane borers and rice borers),
Soazzaanthis pilleriana (grape berry moth), Cydia
peronella (codling moth), Archips spp. (fruit tree
tortrix moths), Plutella xylostella (diamond back
moth); against adults and larvae of Coleoptera
(beetles) e.g. Hypothenemus hampei (coffee berry
borer), Hylesinus spp. (bark beetles), Anthonomus
grandis (cotton boll weevil), Acalymma spp. (cucumber
beetles), Lema spp., Psylliodes spp., Lecinotarsa
decemlineata (Colorado potato beetle), Diabrotica spp.
(corn rootworms), Gonocephalum spp. (false wire
worms), Agriotes spp. (wireworms), Dermolepida and
Heteronychus spp. (white grubs), Phaedon cochleariae
25 (mustard beetle), Lissorhoctus oryzophilus (rice
water weevil), Meligethes spp. (pollen beetles),

Ceutorhynchus spp., Rhynchophorus and Cosmocclites spp. (root weevils); against Hemiptera e.g. Psylla spp., Bemisia spp., Aphis spp., Myzus spp., Mesoura viciae, Phylloxera spp., Adelges spp., Phytocoris humuli (hop damson aphid), Aeneolamia spp., Nephrotettix spp. (rice leaf hoppers), Emcasca spp., Nilaparvata spp., Perkinsiella spp., Pyrilla spp., Aonidiella spp.-(red scales), Coccus spp., Psuedococcus spp., Helopeltis spp. (mosquito bugs), Lyzus spp., Dysdercus spp., Oxycarenus spp., Nezara spp.; Hymenoptera e.g. Athalia spp. and Cephus spp. (saw flies), Atta spp. (leaf cutting ants); Diptera e.g. Hylemyia spp. (root flies), Atherizone spp. and Chlorops spp. (shoot flies), Phytomyza spp. (leaf miners), Ceratitidis spp. (fruit flies); Thysanoptera such as Thrips tabaci; Orthoptera such as Locusta and Schistocerca spp. (locusts) and crickets e.g. Gryllus spp. and Acheta spp.; Collembola e.g. Sminthurus spp. and Onychiurus spp. (springtails), Isoptera e.g. Odontotermes spp. (termites), Dermaptera e.g. Forficula spp. (earwigs) and also other arthropods of agricultural significance such as Acari (mites) e.g. Tetranychus spp., Panonychus spp. and Bryobia spp. (spider mites), Eriophyes spp. (gall mites), Polyphagotarsonemus spp.; Blaniulus spp. (millipedes), Scutigerella spp. (symphiliids), Oniscus spp. (woodlice) and Triops spp. (crustacea); nematodes which attack plants and trees of importance to

agriculture, forestry, horticulture either directly or by spreading bacterial, viral, mycoplasma or, fungal diseases of the plants, root-knot nematodes such as Meloidogyne spp. (e.g. M. incognita); cyst nematodes such as Globodera spp. (e.g. G. rostochiensis); Heterodera spp. (e.g. H. avenae); Radopholus spp. (e.g. R. similis); lesion nematodes such as Pratylenchus spp. (e.g. P. pratensis); Belencladius spp. (e.g. B. gracilis); Tylenchulus spp. (e.g. T. semipenetrans); Rotylenchulus spp. (e.g. R. reniformis); Rotylenchus spp. (e.g. R. robustus); Helicotylenchus spp. (e.g. H. multicinctus); Hemicycliophora spp. (e.g. H. gracilis); Criconemoides spp. (e.g. C. similis); Trichodorus spp. (e.g. T. primitivus); dagger nematodes such as Xiphinema spp. (e.g. X. diversicaudatum), Longidorus spp. (e.g. L. elongatus); Mesolaimus spp. (e.g. M. coronatus); Abelenchoides spp. (e.g. A. ritzenema-bosi, A. besseyi); stem and bulb eelworms such as Ditylenchus spp. (e.g. D. dipsaci).

20 The invention also provides a method for the control of arthropod or nematode pests of plants which comprises the application to the plants or to the medium in which they grow of an effective amount of a compound of general formula I or a pesticidally acceptable salt thereof.

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The compounds of general formula I may be applied in solid or liquid compositions to the soil principally to control those nematodes dwelling therein but also to the foliage principally to control those nematodes attacking the aerial parts of the plants (e.g. Achelenchooides spp. and Ditylenchus spp. listed above).

The compounds of general formula I are of value in controlling pests which feed on parts of the plant remote from the point of application, e.g. leaf feeding insects are killed by the subject compounds applied to roots.

In addition the compounds may reduce attacks on the plant by means of antifeeding or repellent effects.

The compounds of general formula I are of particular value in the protection of field, forage, plantation, glasshouse, orchard and vineyard crops, of ornamentals and of plantation and forest trees, for example, cereals (such as maize, wheat, rice, sorghum), cotton, tobacco, vegetables and salads (such as beans, cole crops, cucurbits, lettuce, onions, tomatoes and peppers), field crops (such as potato, sugar beet, ground nuts, soyabean, oil seed rape), sugar cane, grassland and forage (such as maize, sorghum, lucerne), plantations (such as of tea, coffee, cocoa, banana, oil palm, coconut, rubber,

5 orchards and groves (such as of stone and pip fruit, citrus, kiwifruit, avocado, mango, olives and walnuts), vineyards, ornamental plants, flowers and shrubs under glass and in gardens and parks, forest trees (both deciduous and evergreen) in forests, plantations and nurseries.

10 They are also valuable in the protection of timber (standing, felled, converted, stored or structural) from attack by sawflies (e.g. Urocerus) or beetles (e.g. scolytids, platypodids, lyctids, bostrychids, cerambycids, anobiids).

15 They have applications in the protection of stored products such as grains, fruits, nuts, spices and tobacco, whether whole, milled or compounded into products, from moth, beetle and mite attack. Also protected are stored animal products such as skins, hair, wool and feathers in natural or converted form (e.g. as carpets or textiles) from moth and beetle attack; also stored meat and fish from beetle, mite and fly attack.

20 The compounds of general formula I are of particular value in the control of arthropods or helminths which are injurious to, or spread or act as vectors of diseases in man and domestic animals, for example those hereinbefore mentioned, and more especially in the control of ticks, mites, lice, fleas, midges and biting, nuisance and myiasis

flies. The compounds of general formula I are particularly useful in controlling arthropods or helminths which are present inside domestic host animals or which feed in or on the skin or suck the blood of the animal, for which purpose they may be administered orally, parenterally, percutaneously or topically.

The compositions hereinafter described for topical application to man and animals and in the protection of stored products, household goods, property and areas of the general environment may, in general, alternatively be employed for application to growing crops and crop growing loci and as a seed dressing.

Suitable means of applying the compounds of general formula I include:-

to persons or animals infested by or exposed to infestation by arthropods or helminths, by parenteral, oral or topical application of compositions in which the active ingredient exhibits an immediate and/or prolonged action over a period of time against the arthropods or helminths, for example by incorporation in feed or suitable orally-ingestible pharmaceutical formulations, edible baits, salt licks, dietary supplements, pour-on formulations, sprays, baths, dips, showers, jets, dusts, greases, shampoos, creams, wax-smeats and livestock self-treatment systems;

Environment in general or in specific locations where pests may lurk, including stored products, timber, household goods, and domestic and industrial premises, as sprays, fogs, dusts, smokes, wax-smears, lacquers, granules and baits, and in tricklefeeds to waterways, wells, reservoirs and other running or standing water; to domestic animals in feed to control fly larvae feeding in their faeces.

10 The compounds of general formula I may be applied to control arthropods or helminths in compositions of any type known to the art suitable for internal or external administration to vertebrates or application for the control of arthropods in any premises or indoor or outdoor area, containing as active ingredient at least one compound of general formula I in association with one or more compatible diluents or adjuvants appropriate for the intended use. All such compositions may be prepared in any manner known to the art.

Compositions suitable for administration to vertebrates or man include preparations suitable for oral, parenteral, percutaneous, e.g. pour-on, or topical administration.

Compositions for oral administration comprise one or more of the compounds of general formula I in association with pharmaceutically acceptable carriers or coatings and include, for example, tablets, pills, capsules, pastes, gels, drenches, medicated feeds, medicated drinking water, medicated dietary supplements, slow-release boluses or other slow-release devices intended to be retained within the gastro-intestinal tract. Any of these may incorporate active ingredient contained within microcapsules or coated with acid-labile or alkali-labile or other pharmaceutically acceptable enteric coatings. Feed premixes and concentrates containing compounds of the present invention for use in preparation of medicated diets, drinking water or other materials for consumption by animals may also be used.

Compositions for parenteral administration include solutions, emulsions or suspensions in any suitable pharmaceutically acceptable vehicle and solid or semisolid subcutaneous implants or pellets designed to release active ingredient over a protracted period and may be prepared and made sterile in any appropriate manner known to the art.

Compositions for percutaneous and topical administration include sprays, duses, baths, dips, showers, jects, greases, shampcos, creams, wax-sweats, or pour-on preparations and devices (e.g. ear tags) 5 attached externally to animals in such a way as to provide local or systemic arthropod control.

Solid or liquid baits suitable for controlling arthropods comprise one or more compounds of general formula I and a carrier or diluent which may include a 10 food substance or some other substance to induce consumption by the arthropod.

Liquid compositions include water miscible 15 concentrates, emulsifiable concentrates, flowable suspensions, wettable or soluble powders containing one or more compounds of general formula I which may be used to treat substrates or sites infested or liable to infestation by arthropods including 20 premises, outdoor or indoor storage or processing areas, containers or equipment and standing or running water.

Solid homogenous or heterogenous compositions 25 containing one or more compounds of general formula I, for example granules, pellets, briquettes or capsules, may be used to treat standing or running water over a period of time. A similar effect may be achieved using trickle or intermittent feeds of water dispersible concentrates as described herein.

Compositions in the form of aerosols and aqueous or non-aqueous solutions or dispersions suitable for spraying, fogging and low- or ultra-low volume spraying may also be used.

5 Suitable solid diluents which may be used in the preparation of compositions suitable for applying the compounds of general formula I include aluminium silicate, kieselguhr, corn husks, tricalcium phosphate, powdered cork, adsorbent carbon black, 10 magnesium silicate, a clay such as kaolin, bentonite or attapulgite, and water soluble polymers and such solid compositions may, if desired, contain one or more compatible wetting, dispersing, emulsifying or colouring agents which, when solid, may also serve as 15 diluent.

Such solid compositions, which may take the form of dusts, granules or wettable powders, are generally prepared by impregnating the solid diluents with solutions of the compound of general formula I in 20 volatile solvents, evaporating the solvents and, if necessary, grinding the products so as to obtain powders and, if desired, granulating or compacting the products so as to obtain granules, pellets or briquettes or by encapsulating finely divided active 25 ingredient in natural or synthetic polymers, e.g. gelatin, synthetic resins and polyamides.

The wetting, dispersing and emulsifying agents which may be present, particularly in wettable powders, may be of the ionic or non-ionic types, for example sulphoricinoleates, quaternary ammonium derivatives or products based upon condensates of ethylene oxide with nonyl- and octyl-phenol, or carboxylic acid esters of anhydrosorbitols which have been rendered soluble by etherification of the free hydroxy groups by condensation with ethylene oxide, or mixtures of these types of agents. Wettable powders may be treated with water immediately before use to give suspensions ready for application.

Liquid compositions for the application of the compounds of general formula I may take the form of solutions, suspensions and emulsions of the compounds of general formula I optionally encapsulated in natural or synthetic polymers, and may, if desired, incorporate wetting, dispersing or emulsifying agents. These emulsions, suspensions and solutions may be prepared using aqueous, organic or aqueous-organic diluents, for example acetophenone, isophotone, toluene, xylene, mineral, animal or vegetable oils, and water soluble polymers (and mixtures of these diluents), which may contain wetting, dispersing or emulsifying agents of the ionic or non-ionic types or mixtures thereof, for example

those of the types described above. When desired, the emulsions containing the compounds of general formula I may be used in the form of self-emulsifying concentrates containing the active substance dissolved in the emulsifying agents or in solvents containing emulsifying agents compatible with the active substance, the simple addition of water to such concentrates producing compositions ready-for use. -----

Compositions containing compounds of general formula I which may be applied to control arthropod, plant nematode or helminth pests, may also contain synergists (e.g. piperonyl butoxide or sesame), stabilizing substances, other insecticides, acaricides, plant nematocides or anthelmintics, fungicides (agricultural or veterinary as appropriate e.g. benomyl, iprodione), bactericides, arthropod or vertebrate attractants or repellants or pheromones, reodorants, flavouring agents, dyes and auxiliary therapeutic agents, e.g. trace elements. These may be designed to improve potency, persistence, safety, uptake where desired, spectrum of pests controlled or to enable the composition to perform other useful functions in the same animal or area treated.

Examples of other pesticidally-active compounds which may be included in, or used in conjunction with, the compositions of the present invention are:-
5 acephate, chlorpyrifos, demeton-S-methyl, disulfoton, ethopropofos, fenitrothion, malathion, monocrotophos, parathion, phosalone, pirimiphos-methyl, triazophos, cyfluthrin, cypermethrin, deltamethrin, fenpropathrin, fenvaerate, permethrin, aldicarb, carbosulfan, methomyl, oxamyl, pirimicarb, bendiocarb,
10 teflubenzuron, dicofol, endosulfan, lindane, benzoximate, carkap, cyhexatin, tetradifon, avermectins, ivermectin, milbemycins, thiophanate, 2-chlorfon and dichlorvos.

15 The compositions for application to control arthropods usually contain from 0.00001% to 95%, more particularly from 0.0005% to 50%, by weight of one or more compounds of general formula I or of total active ingredients (that is to say the compound(s) of general formula I together with other substances toxic to arthropods and plant nematodes, anthelmintics,
20 synergists, trace elements or stabilisers). The actual compositions employed and their rate of application will be selected to achieve the desired effect(s) by the farmer, livestock producer, medical or veterinary practitioner, pest control operator or
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other person skilled in the art. Solid and liquid compositions for application topically to animals, timber, stored products or household goods usually contain from 0.00005% to 90%, more particularly from 5 0.001% to 10%, by weight of one or more compounds of general formula I. For administration to animals orally or parenterally,—including percutaneously, solid and liquid compositions normally contain from 0.1% to 10 90% by weight of one or more compounds of general formula I. Medicated feedstuffs normally contain from 0.001% to 3% by weight of one or more compounds of general formula I. Concentrates and supplements for mixing with feedstuffs normally contain from 5% to 15 90%, and preferably from 5% to 50%, by weight of one or more compounds of general formula I. Mineral salt licks normally contain from 0.1% to 10% by weight of 20 one or more compounds of general formula I.

Dusts and liquid compositions for application to 25 livestock, persons, goods, premises or outdoor areas may contain 0.0001% to 15%, and more especially 0.005% to 2.0%, by weight of one or more compounds of general formula I. Suitable concentrations in treated waters are between 0.0001 ppm and 20 ppm, and more especially 0.001 ppm to 5.0 ppm, of one or more compounds of general formula I and may also be used therapeutically 25 in fish farming with appropriate exposure times.

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Edible baits may contain from 0.01% to 5% and preferably 0.01% to 1.0%, by weight of one or more compounds of general formula I.

When administered to vertebrates parenterally, orally or by percutaneous or other means, the dosage of compounds of general formula I will depend upon the species, age and health of the vertebrate and upon the nature and degree of its actual or potential infestation by arthropods. A single dose of 0.1 to 10 100 mg, preferably 2.0 to 20.0 mg, per kg body weight of the animal or doses of 0.01 to 20.0 mg, preferably 0.1 to 5.0 mg, per kg body weight of the animal per day for sustained medication are generally suitable by oral or parenteral administration. By use of 15 sustained release formulations or devices, the daily doses required over a period of months may be combined and administered to animals on a single occasion.

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The present invention accordingly provides an arthropodical, plant nematocidal or antihelmintic composition which comprises at least one compound of general formula I, or a salt thereof, in association with one or more compatible diluents or carriers with the provisos that (1) when the composition comprises a single compound of general formula I wherein R^4 and Z both represent methyl, Y represents thiocyanato and $(R^3)_n$ represents 2-, 3- or 4-nitro, 4-methyl, 10 4-chloro or 2,4-dinitro substitution; or R^4 represents methyl, Y represents cyano, Z represents unsubstituted amino and $(R^3)_n$ represents 4-chloro, 2,4-dichloro, 3,4-dichloro, 3-chloro-4-methyl or 2-methyl-4-chloro substitution, the composition is not an association of a single compound of general formula I alone with water or a common organic solvent; (2) when the composition comprises a single compound of general formula I wherein R^4 represents methyl, Y represents cyano or $CONH_2$, Z represents unsubstituted amino and $(R^3)_n$ represents 3- or 20 4-fluoro substitution; or R^4 represents ethyl, Y represents cyano or $CONH_2$, Z represents unsubstituted amino and $(R^3)_n$ represents 3- or 4-chloro, 2-, 3- or 4-fluoro or methyl, 3-bromo or 4-nitro substitution; or R^4 represents propyl, Y represents cyano or $CONH_2$, Z represents unsubstituted amino and $(R^3)_n$ represents 3-fluoro 25 4-nitro substitution.

substitution; or R⁴ represents methyl, Y represents sulphamoyl, Z represents chloro and (R³)_n represents 4-chloro substitution; the composition comprises an agriculturally acceptable surface active agent or a feedstuff; (3) when R⁴ represents methyl, Y represents nitro, and Z represents chloro or R⁴ represents chloro, Y represents nitro, and Z represents methyl and (R³)_n represents 4-nitro, the composition comprises a pharmaceutically acceptable adjuvant or a feedstuff or is substantially sterile and pyrogen-free or is in unit dosage form; and (4) excluding compositions comprising 1-(4-nitrophenyl)-3-nitro-4-pyrazole-carbonitrile or carboxamide.

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Medicated feeds which comprise known compounds of general formula I and arthropodicidally- or anhelminthically-acceptable salts thereof and an edible carrier or diluent form a feature of the present invention.

In experiments on activity against arthropods carried out on representative compounds, the following results (wherein 'Dose mg/kg' indicates the dose of test compound administered in mg per kg animal body weight and ppm indicates the concentration of the compound in parts per million of the test solution applied) have been obtained:-

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Test 1

One or more dilutions of the compounds to be tested were made in 50% aqueous acetone.

5 a) Test species: Plutella xylostella (Diamond-back Moth) and Phaedon cochleariae (Mustard Beetle).

10 Turnip leaf discs were set in agar in petri-dishes and infected with 10 larvae (2nd instar Plutella or 3rd instar Phaedon). Four replicate dishes were assigned to each treatment and were sprayed under a Potter Tower with the appropriate test dilution. Four or five days after treatment the dishes were removed from the constant temperature (25°C) room in which they had been held and the mean percentage mortalities of larvae were determined. These data 15 were corrected against the mortalities in dishes treated with 50% aqueous acetone alone which served as controls.

b) Mesoura viciae (Vetch Aphid)

20 Potted tic bean plants previously infected with mixed stages of Mesoura were sprayed to run-off using a laboratory turntable sprayer. Treated plants were held in a greenhouse for 2 days and were assessed for aphid mortality using a scoring system, judging the response in comparison with plants treated with 50% aqueous acetone alone, as controls.

Score

3	all aphids dead
2	few aphids alive
1	most aphids alive
5	0 no significant mortality

According to the above method (a) an application of 500 ppm of the following compounds was totally effective against the larvae of Plutella xylostella, producing 100% mortality.

10 Compound No.

5, 6, 7, 8, 20, 21, 22, 28, 30, 31, 32, 35, 36, 37, 38, 39, 41, 42, 43, 44, 68, 69, 70, 71, 72, 73, 76, 79, 80, 81, 85, 87, 94, 99, 102, 103, 104, 105, 106, 108, 111, 120, 121

15 According to the above method (a) an application of 5 ppm of the following compounds was totally effective against the larvae of Phaedon cochleariae, producing 100% mortality.

Compound No.

20 36, 53, 57, 58, 70, 71, 74, 79, 80, 85, 90, 91, 97, 98, 99, 102, 104, 106, 108, 109, 111, 112, 113, 116, 118, 120, 121

According to the above method (b) an application of 50 ppm of the following compounds was totally effective against Megoura viciae producing 100% mortality, that is giving a score of 12 from 4 replicates.

Compound No

4, 5, 20, 21, 36, 48, 53, 57, 58, 82, 83, 92, 93,
98, 102, 106, 109, 111, 116, 117, 118, 120

The data quoted in Tables 1-3 summarise the results from a number of different experiments carried out to the protocols a) and b) above.

Table 1

	<u>No.</u>	Plutella Zm 500 ppm	Phaedon Zm 10 ppm	Megoura score/12 50 ppm	
15	5		100		
	6		100	9	13
	7		100	10	
	19	100*	100	10	
	20		100		
20	21		100		
	42		100	10	20
	10	73	45	10	
	43		100	11	

Table 2

<u>No.</u>	<i>Plutella</i> 500 ppm	<i>Phaedon</i> 10 ppm
8		93
2	89	100
47	96	100
35		100
22		100
23	58	100
24	10	100
30		100
28		56
29	48	100
25	21*	100
31		16
38		85
41		100
37		100
33	44	100
44		100
32		84
40	98	21
39		100

Table 3

No.	Phaeden Zn	10 ppm	Megouta score/12	50 ppm
5	4	98		
	45	100		11
	9	68		10+

* % mortality at 100 ppm

+ score at 10 ppm

10

Test 2

Some 20 larvae of Rhizicephalus appendiculatus were placed in plastic capsules attached to the shaved flank of guinea-pigs. After 3 hours and then at 23 hourly intervals, the guinea-pigs were given a total of 4 subcutaneous injections of the test compound. Approximately 100 hours after infestation, the guinea-pigs were killed and the engorged tick larvae recovered, counted and kept at 23°C in a humidity cabinet for 14 to 21 days. After this period, the percentage survival through moulting was assessed. Results obtained are given below in Table 4.

Table 4

Compound No.	Dose (mg/kg at each repetition)	Result
5	5	No ticks recovered
	4	No ticks recovered
	12	Less than five engorged ticks recovered
	2.5	Ticks engorged normally but only 62.5 per cent survived
10	5	No ticks recovered
	4	No ticks recovered
	3	Number of engorged ticks reduced, only 8.3 per cent survived
	5	Ticks engorged normally but only 30.0 per cent survived
15	2.5	Ticks engorged normally but only 47.9 per cent survived
	10	No ticks recovered
	26	Less than five engorged ticks recovered
	5	No ticks recovered
20	15	No ticks recovered
	25	No ticks recovered
	2.5	No ticks recovered

20

25

Test 3

The high activity of the compounds of general formula against the cockroach species Periplaneta americana is demonstrated by results from the 5 following experiment.

0.2 microlitres of an acetone solution of the compound was injected through the soft cuticle between the leg and thorax of ten insects, to give a dose rate of 5 micrograms per g of insect body weight. Ten 10 cockroaches were similarly injected with 0.2 microlitres of acetone alone to serve as controls. After treatment the insects were held in plastic boxes with appropriate food. Five days after treatment the numbers of dead and alive insects were counted and 15 percentage mortalities calculated.

According to the above method a dose of 5 micrograms/g insect body weight of the following compounds was totally effective against the cockroach species Periplaneta americana producing 100% mortality.

20 Compound No.

2, 5, 14, 17, 22, 53

The following Examples illustrate compositions for use against arthropod, plant nematode or helminth pests which comprise, as active ingredient, compounds 25- of general formula I.

Example A

A dusting powder may be prepared by intimately mixing:-

5-amino-4-cyano-1-(2,6-dichloro-4-

5 trifluoromethylphenyl)- 1 to 10% w/w
3-trifluoromethylpyrazole. (weight/weight)

Talc superfine to 100% by weight

This powder may be applied to a locus of arthropod infestation, for example refuse tips or dumps, stored products or household goods or animals infested by, or at risk of infestation by, arthropods to control the arthropods by oral ingestion. Suitable means for distributing the dusting powder to the locus of arthropod infestation include mechanical blowers, handshakers and livestock self treatment devices.

The 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole may, if desired, be replaced in the above dusting powder by any other compound of general formula I.

Example B

An edible bait may be prepared by intimately mixing:-

5-amino-4-cyano-1-(2,6-dichloro-4-

trifluoromethylphenyl)- 0.1 to 1.0% w/w
3-trifluoromethylpyrazole

Wheat flour 80% w/w

Molasses to 100% w/w

This edible bait may be distributed at a locus, for example domestic and industrial premises, e.g. kitchens, hospitals or stores, or outdoor areas, infested by arthropods, for example ants, locusts, cockroaches and flies, to control the arthropods by oral ingestion.

5 The 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole may, if desired, be replaced in the above edible bait by any 10 other compound of general formula I.

Example C

A solution may be prepared containing:-

5-amino-4-cyano-1-(2,6-dichloro-4-

15 trifluoromethylphenyl)-

..... 15% w/v

3-trifluoromethylpyrazole

(weight/volume)

Dimethylsulphoxide to 100% by volume by dissolving the pyrazole derivative in a portion of the dimethylsulphoxide and then adding more dimethylsulphoxide to the desired volume. This solution may

20 be applied to domestic animals infested by arthropods, percutaneously as a pour-on application or, after sterilisation by filtration through a polytetrafluoroethylene membrane (0.22 μm pore size), by parenteral injection, at a rate of application of from 1.2 to 25 12 ml of solution per 100 kg of animal body weight.

The 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole may, if desired, be replaced in the above solution by similar amounts of any other compound of general formula I.

5

Example D

A wettable powder may be formed from:-

5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole 50% w/w
10 Ethylan BCP (a nonylphenol/ethylene oxide condensate containing 9 moles of ethylene oxide per mol of phenol) 5% w/w
Aerosil (silicon dioxide of microfine-particle size) 5% w/w
15 Celite PF (synthetic magnesium silicate carrier) 40% w/w
by adsorbing the Ethylan BCP onto the Aerosil, mixing with the other ingredients and grinding the mixture in a hammer-mill to give a wettable powder, which may be
20 diluted with water to a concentration of from 0.001% to 2% w/v of the pyrazole compound and applied to a locus of infestation by arthropods, for example dipterous larvae, or plant nematodes by spraying, or to domestic animals infested by, or at risk of infestation by, arthropods, by spraying or dipping, or
25 by oral administration as drinking water, to control the arthropods or helminths.

The 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoro-methylphenyl)-3-trifluoromethylpyrazole may, if desired, be replaced in the above wettable powder by any other compound of general formula I.

5

Example E

A slow release bolus may be formed from granules containing a density agent, binder, slow-release agent and 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoro-methylphenyl)-3-trifluoromethylpyrazole compound at varying percentage compositions. By compressing the mixture a bolus with a specific gravity of 2 or more can be formed and may be administered orally to ruminant domestic animals for retention within the reticulo-rumen to give a continual slow release of pyrazole compound over an extended period of time to control infestation of the ruminant domestic animals by arthropods or helminths.

The 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoro-methylphenyl)-3-trifluoromethylpyrazole may, if desired, be replaced in the above bolus by any other compound of general formula I.

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Example F

A slow release-composition may be prepared from:-

5-amino-4-cyano-1-(2,6-dichloro-4-

trifluoromethylphenyl)- 0.5 to 25% w/w

5 3-trifluoromethylpyrazole

polyvinylchloride base to 100% w/w

by blending the polyvinylchloride base with the

pyrazole compound and a suitable plasticiser, e.g.

dioctyl phthalate, and melt-extruding or hot-moulding

10 the homogenous composition into suitable shapes, e.g.

granules, pellets, briquettes or strips, suitable, for

example, for addition to standing water or, in the

case of strips, fabrication into collars or ear-tags

for attachment to domestic animals, to control insect

15 pests by slow release of the pyrazole compound.

20

25

The compounds of general formula I can be prepared by the application or adaptation of known methods (i.e. methods heretofore used or described in the chemical literature), generally heterocycle formation followed where necessary by changing substituents with protection/deprotection of other substituents if necessary, for example as hereinafter described.

In the following description when symbols appearing in formulae are not specifically defined it is to be understood that they are "as hereinbefore defined" in accordance with the first definition of each symbol in this specification.

Compounds of general formula I conforming to general formula IA wherein Y' represents the cyano or nitro group or a group RSO_2 , RSO or RS , a straight- or branched-chain alkoxy carbonyl group containing from 2 to 7 carbon atoms, or a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms, Z' represents the unsubstituted amino group or a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms, and R⁵ represents a fluorine, chlorine or bromine atom, the cyano group or a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms which may be unsubstituted or substituted by one or more halogen

atoms, or a cycloalkyl group containing from 3 to 6 carbon atoms, may be prepared by the process which comprises

(i) the reaction of a compound of the general formula II, or an acid addition salt thereof, e.g. the hydrochloride, with (1), when R⁵ in the compound of general formula IA represents a fluorine, chlorine or bromine atom, an optionally halogenated straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms or a cycloalkyl group containing from 3 to 6 carbon atoms, a compound of the general formula III, wherein R⁶ represents the cyano group or a straight- or branched-chain alkanoyl group containing from 2 to 5 carbon atoms and R⁸ represents a straight- or branched-chain alkoxy group containing from 1 to 4 carbon atoms, preferably ethoxy, the hydroxy group or a fluorine, chlorine or bromine atom, or (2), when R⁵ in the compound of general formula IA represents the cyano group (and Y' represents the cyano group and Z' represents the unsubstituted amino group), tetracyanoethylene.

The reaction of a compound of general formula II with a compound of general formula III (optionally prepared in situ) or tetracyanoethylene may be effected in the presence of an inert organic solvent, for example an alkanol containing from 1 to 4 carbon atoms, e.g. ethanol, acetic acid, ethoxyethanol or an ether, and at a temperature from ambient temperature to the reflux temperature of the reaction mixture and optionally in the presence of an alkali metal, e.g. sodium or potassium, acetate, carbonate or bicarbonate or organic base e.g. triethylamine. When an acid addition salt of the compound of general formula II is used, the reaction with the compound of general formula III is effected in the presence of an alkali metal, e.g. sodium or potassium, acetate, carbonate or bicarbonate.

(ii) Compounds of general formula IA wherein Z' represents the unsubstituted amino group may alternatively be prepared directly by reacting a compound of general formula $Y'CH_2CN$ with a compound of general formula $R^7C(R^0)_3$ wherein R^7 represents a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms or a cycloalkyl group containing from 3 to 6 carbon atoms and R^0 represents an alkoxy group which may be

straight- or branched-chain and preferably contains from 1 to 4 carbon atoms, in an inert organic solvent, preferably ethanol, at a temperature from ambient to reflux.

5 (iii) Compounds of general formula IA wherein Z' represents the unsubstituted amino group and R⁵ represents the cyano group may be obtained by the reaction of a compound of the general formula IV with a molar equivalent of compound of general formula
10 Y'CH₂CN, i.e. malononitrile when Y' represents the cyano group, generally in the presence of an anhydrous inert organic solvent, e.g. ethanol, and a molar equivalent of a base, e.g. sodium hydride, and at a temperature from 0 to 50°C.

15 The compounds of general formula IA may be prepared by reaction of a compound of general formula II with a compound of general formula III or tetracyanoethylene with isolation of an intermediate compound of general formula V from the reaction
20 mixture. When the reaction of a compound of general formula II with a compound of general formula III is effected in acetic acid, in the absence or presence of an alkali metal, e.g. sodium or potassium, acetate, the intermediate compound of general formula V may
25 separate from the reaction mixture, depending upon its solubility in the reaction medium, and may, if desired, be isolated before being cyclised as

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hereinbefore described to a compound of general formula IA. The cyclisation of a compound of general formula V, which constitutes a feature of the invention may be effected in the presence of an inert organic solvent, for example an alcohol containing from 1 to 4 carbon atoms, e.g. ethanol, acetic acid or ethoxyethanol, at a temperature of from ambient temperature up to the reflux temperature of the reaction mixture, and optionally in the presence of sodium ethoxide when the solvent is ethanol.

It will be appreciated that in the preparation of compounds of general formula I the following subsidiary processes or adaptations thereof may be performed in an appropriate combination to achieve the compound sought.

Compounds of general formula I which conform to general formula IB wherein R¹ represents an R⁹C(=O)- group, wherein R⁹ represents a straight- or branched-chain alkyl or alkoxy group containing from 1 to 6 carbon atoms, or a cycloalkyl group containing from 3 to 6 carbon atoms and R² represents a hydrogen atom or an R⁹C(=O)- group which is identical to the group R⁹C(=O)- represented by R¹ or -NR¹R² represents a cyclic imide as hereinbefore defined, may be prepared by the reaction

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of a compound of general formula I wherein Z represents the unsubstituted amino group, or an alkali metal salt thereof, with a compound of the general formula:-



wherein X represents a chlorine or bromine atom, or with a compound of the general formula:-



or with a dicarboxylic acid derivative. The reaction may be conducted in the absence or presence of an inert organic solvent, for example acetonitrile, tetrahydrofuran, a ketone, e.g. acetone, an aromatic hydrocarbon, e.g. benzene or toluene, chloroform, dichloromethane or dimethylformamide, and optionally in the presence of an acid-binding agent, for example pyridine, triethylamine or an alkali metal, e.g. sodium or potassium, carbonate or bicarbonate, at a temperature from 0°C to the reflux temperature of the reaction medium, to give a compound of general formula I B wherein R¹ represents an R⁹C(-J)- group wherein R⁹ is as hereinbefore defined and R² represents a hydrogen atom or an R⁹C(-O)- group, depending upon the reaction conditions chosen and/or the use of an excess of the compound of general formula VI or VII.

Compounds of general formula IB wherein R¹ represents a formyl group and R² represents a hydrogen atom may be prepared by reaction of a compound of general formula I, wherein Z represents the unsubstituted amino group with formic acid. The reaction may be conducted in an inert organic solvent, for example a ketone e.g. methylisobutyl ketone, or an aromatic hydrocarbon, e.g. benzene or toluene, at the reflux temperature of the reaction mixture.

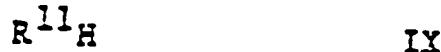
Compounds of general formula IB wherein R¹ represents a formyl group and R² represents a hydrogen atom or a formyl group, may be prepared by the reaction of a compound of general formula I, wherein Z represents the unsubstituted amino group with formylacetic anhydride. Formylacetic anhydride may be prepared from formic acid and acetic anhydride and the reaction with the compound of general formula I may be conducted in the absence or presence of an inert organic solvent, for example a ketone, e.g. acetone, or an aromatic hydrocarbon, e.g. benzene or toluene, and optionally in the presence of an acid-binding agent, for example pyridine, triethylamine or an alkali metal, e.g. sodium or potassium, carbonate or bicarbonate, at a temperature from 0°C to the reflux temperature of the reaction mixture, to give a compound of general formula IB

wherein R^1 represents a formyl group and R^2 represents a hydrogen atom or a formyl group, depending upon the reaction conditions chosen and/or the use of an excess of formylacetic anhydride.

5 Compounds of general formula IB wherein R^1 represents a formyl group or a group $R^9C(-O)-$ and R^2 represents a hydrogen atom may be prepared by the selective removal by hydrolysis of an $R^9C(-O)-$ group or a formyl group from a compound of general formula 10, IB wherein R^1 and R^2 both represent a $R^9C(-O)$ group or a formyl group. Hydrolysis is effected under mild conditions, for example by treatment with an aqueous-ethanolic solution or suspension of an alkali metal, e.g. sodium or potassium, bicarbonate, 15 or with aqueous ammonia.

Compounds of general formula IB wherein R^1 represents a straight- or branched-chain 20 alkoxy carbonyl group containing from 2 to 7 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, and R^2 represents a hydrogen atom may be prepared by the reaction of a compound of the general formula VIII, wherein R^{10} represents an alkoxy carbonyl group $R^{11}C(-O)$, wherein R^{11} represents a straight- or branched-chain alkoxy group 25 containing from 1 to 6 carbon atoms (which is unsubstituted or substituted by one or more halogen atoms) or a phenoxy group, with a compound of the

general formula:-



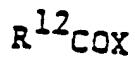
to replace a first group represented by the symbol R^{10} by a hydrogen atom, and to replace the second group represented by the symbol R^{10} by an alkoxy carbonyl group when R^{10} represents a phenoxy carbonyl group, or, if desired, to replace the second group represented by the symbol R^{10} by another alkoxy carbonyl group when R^{10} in formula VIII represents an alkoxy carbonyl group. As will be apparent to those skilled in the art, the desired compound of general formula IX is obtained by selection of the appropriate compounds of general formulae VIII and IX. The reaction may be effected in water or an inert aqueous-organic or organic solvent, for example an alkanol containing 1 to 4 carbon atoms, e.g. ethanol, or an aromatic hydrocarbon, e.g. benzene or toluene, or which is preferably an excess of the compound of general formula IX, at a temperature from ambient temperature to the reflux temperature of the reaction mixture and, if necessary, at elevated pressure, and optionally in the presence of a base, for example an alkali metal alkoxide, e.g. of the compound of general formula IX.

Compounds of general formula IB wherein R¹ and R², which may be the same or different, each represents a formyl group or a R⁹C(=O)- group, may be prepared by the reaction of an alkali metal, e.g. sodium or potassium, derivative of a compound of general formula IB wherein R¹ represents a group R⁹C(=O)- as hereinbefore defined, for a formyl group, and R² represents a hydrogen atom with formic acid, formylacetic anhydride or a compound of general formula VI. Reaction may be effected in an inert aprotic solvent, e.g. dimethylformamide, at a temperature from laboratory temperature to the reflux temperature of the reaction mixture.

Alkali metal derivatives of compounds of general formula I (wherein Z represents the unsubstituted amino group) or IB wherein R¹ represents a group R⁹C(=O)- and R² represents a hydrogen atom may be prepared in situ by reaction with an alkali metal, e.g. sodium or potassium, hydride, in an inert aprotic solvent, e.g. dimethylformamide, at a temperature from laboratory temperature to the reflux temperature of the reaction mixture.

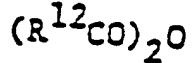
Compounds of general formula VIII wherein R¹⁰ represents a group R¹¹C(=O)-, may be prepared as hereinbefore described. Compounds of general formula VIII wherein R¹⁰ represents a phenoxy carbonyl group may be prepared by the reaction of a compound of

general formula I (wherein Z represents the unsubstituted amino group), with a compound of the general formula:-



VIA

5 , wherein R^{12} represents a phenoxy group, or with a compound of the general formula:-



VIIA

10 using the reaction conditions hereinbefore described for the reaction of a compound of general formula I with a compound of formula VI or VII.

15 Compounds of general formula IB wherein R^1 represents a group R^{13} which represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms (which may be unsubstituted or substituted by alkoxy carbonyl groups containing from 2 to 5 carbon atoms) or a cycloalkyl group containing from 3 to 6 carbon atoms, and R^2 represents a hydrogen atom may be prepared by the removal of the group $R^9C(-O)-$ of a compound of the general formula IB, wherein R^1 represents a group R^{13} and R^2 represents a group $R^9C(-O)-$. Removal of the group $R^9C(-O)-$ may be effected by selective hydrolysis under mild conditions, for example by treatment with an alkali metal, e.g. sodium or potassium, hydroxide in water or an inert organic or aqueous-organic solvent, for example a lower alkanol, e.g. methanol, or a mixture of water and lower alkanol, at a

temperature from laboratory temperature up to the reflux temperature of the reaction mixture.

Compounds of general formula IB, wherein R^1 represents a group R^{13} and R^2 represents a group $R^9C(-O)-$, may be prepared by reaction of a compound of general formula IB wherein R^1 represents a hydrogen atom, or an alkali metal, e.g., sodium or potassium, derivative thereof, with a compound of the general formula:-



, wherein X^1 represents a chlorine, bromine or iodine atom. Reaction may be effected in an inert organic solvent, e.g. dichloromethane, tetrahydrofuran, or dimethylformamide, at a temperature from laboratory temperature up to the reflux temperature of the reaction mixture and, when a compound of general formula IB is used, in the presence of a base, e.g. Triton B; or by reaction of a compound of general formula IB wherein R^1 represents the hydrogen atom and R^2 represents a group R^{13} with a compound of general formula VI or VII.

Compounds of general formula I wherein Z represents an N -(alkyl or cycloalkyl)- N -formylamino group as hereinbefore described may be prepared in a similar manner to the process above using, where appropriate, formylacetic anhydride instead of a compound of general formula VI or VII.

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Compounds of general formula I3 wherein one of R¹ and R² or both of R¹ and R² represent a straight- or branched- chain alkyl group containing from 1 to 6 carbon atoms or cycloalkyl group containing from 3 to 6 carbon atoms, groups represented by R¹ and R² being identical, may be prepared by reaction of a compound of general formula I, wherein Z represents the unsubstituted amino group, or an alkali metal, e.g. sodium or potassium, derivative thereof, with a compound of general formula X, in the absence or presence of an inert organic solvent, for example an aromatic hydrocarbon, e.g. benzene or toluene, chloroform, dichloromethane, tetrahydrofuran or dimethylformamide, and optionally in the presence of an acid-binding agent, for example pyridine, triethylamine or an alkali metal, e.g. sodium or potassium, bicarbonate, at a temperature from 0°C up to the reflux temperature of the reaction mixture.

Alkali metal derivatives of compounds of formulae I3 (wherein R¹ represents a hydrogen atom) and I (wherein Z represents the unsubstituted amino group) may be prepared in situ by the reaction of the compounds, with an alkali metal, e.g. sodium or potassium, hydride, at a temperature from laboratory temperature to the reflux temperature of the reaction mixture.

Compounds of general formula I wherein Z represents a straight- or branched-chain alkoxyethyleneamino group containing from 2 to 5 carbon atoms which may be unsubstituted or substituted on methylene by a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms may be prepared by the reaction of a compound of general formula I (wherein Z represents the unsubstituted amino group) with a trisalkoxyalkane in the presence of an acidic catalyst, e.g. p-toluenesulphonic acid, at a temperature from ambient temperature to the reflux temperature of the reaction mixture.

Compounds of general formula I, wherein Z represents a straight- or branched-chain alkylsulphenylamino group containing from 1 to 4 carbon atoms, may be prepared by the reaction of compounds of general formula I (wherein Z represents the unsubstituted amino group) with an alkanesulphenyl chloride in the presence of a base, e.g. sodium hydride, and optionally in the presence of a crown ether catalyst, e.g. 15-crown-5.

The reaction may be performed in a solvent, e.g. tetrahydrofuran, at a temperature from 0°C to the reflux temperature of the reaction mixture.

Compounds of general formula I wherein Z represents $-\text{NHCH}_2\text{R}^{14}$ wherein R^{14} represents the hydrogen atom or a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms may be
5 prepared by reaction of a compound of general formula I wherein Z represents $-\text{N}=\text{C}(\text{OR}^{15})\text{R}^{14}$ wherein R^{15} represents a straight- or branched-chain alkyl-group containing from 1 to 4 carbon atoms with a reducing agent, preferably sodium borohydride. The reaction
10 may be effected in an inert organic solvent, ethanol or methanol being preferred, at a temperature from 0°C to the reflux temperature of the reaction mixture.

Compounds of general formula I wherein Y represents $-\text{C}(=\text{O})\text{NH}_2$ may be prepared by partial
15 hydrolysis of a compound of general formula I wherein Y represents $-\text{CN}$ preferably with sulphuric acid at a temperature from ambient temperature to 100°C.

Compounds of general formula I wherein Y represents the chlorine, bromine or iodine atom may be
20 prepared by reaction of a compound of general formula XI with a halogenating agent, preferably N-halosuccinimide in an inert solvent, preferably carbon tetrachloride, at a temperature from ambient temperature to the reflux temperature of the reaction
25 mixture.

Compounds of general formula I wherein Z
represents the chlorine, bromine or iodine atom may be
prepared by diazocisation of a compound of general
formula I wherein Z represents $-\text{NH}_2$ with an alkyl
5 nitrite, preferably tert-butyl nitrite, in the
presence of a halogenating agent preferably bromoform,
iodine or anhydrous copper chloride at a temperature
from 0°C to 100°C .

Compounds of general formula I wherein Y
10 represents the nitro group may be prepared by reacting
a compound of general formula XI with a nitrating
agent, preferably nitric acid optionally in the
presence of sulphuric acid or nitric acid in a
solvent such as acetic acid or acetic anhydride at a
15 temperature from 0°C to 100°C .

Compounds of general formula I wherein Y
represents $-\text{SO}_2\text{NR}^{16}\text{R}^{17}$ wherein R^{16} and R^{17} ,
which may be the same or different, each represent the
hydrogen atom or a straight- or branched-chain alkyl
20 group containing from 1 to 6 carbon atoms may be
prepared by reacting a compound of general formula XIV
with an amine of the general formula $\text{R}^{16}\text{R}^{17}\text{NH}$ in a
solvent such as toluene or water at a temperature from
 0°C to the reflux temperature of the reaction mixture.

Compounds of general formula I wherein Y represents $-\text{CONR}^{16}\text{R}^{17}$ may be prepared by reacting a compound of general formula XV wherein X^2 represents a chlorine or bromine atom or activated ester moiety e.g. 4-nitrophenoxy group, especially the chlorine atom, with an amine of the general formula $\text{R}^{16}\text{R}^{17}\text{NH}$, in a solvent such as toluene or water, at a temperature from 0°C to the reflux temperature of the reaction mixture.

10 Intermediates of general formula XI may be prepared by decarboxylation of a compound of general formula XVI, performed by heating at a temperature from 100°C to 250°C optionally in the presence of an inert organic solvent, particularly 15 N,N-dimethylaniline.

15 Intermediates of general formula XI wherein Z is the unsubstituted amino group and R^4 represents a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms which may be unsubstituted or 20 substituted by one or more halogen atoms may also be prepared by reaction of an appropriate β -ketonitrile or derivative thereof e.g. the imine with an arylhydrazine in an inert organic solvent such as ethanol optionally in the presence of an acidic or 25 basic catalyst at a temperature from ambient to 100°C.

Alternatively intermediates of general formula XI
may be prepared directly from esters of compounds of
general formula XVI by heating in an inert organic
solvent preferably acetic acid at a temperature from
5 50°C to reflux, in the presence of a strong acid
preferably hydrobromic acid.

Intermediates of general formula XVI may be
prepared by hydrolysis of esters of general formula I
wherein Y represents $-COOR^{18}$ wherein R^{18}
10 represents a straight- or branched- chain alkyl group
containing from 1 to 6 carbon atoms, preferably with
an alkali metal hydroxide in a solvent such as an
aqueous alcohol at a temperature from 0°C to the
reflux temperature of the reaction mixture.

15 Intermediates of general formula XIV may be
prepared by reacting a compound of general formula XI
with chlorosulphonic acid at a temperature from 0°C to
150°C.

20 Intermediates of general formula XV are prepared
by reacting a compound of general formula XVI with a
chlorinating or brominating agent or e.g.
4-nitrophenol (preferably thionyl chloride) at a
temperature from ambient temperature to the reflux
temperature of the reaction mixture.

Compounds of general formula I wherein Y represents $-C(=O)R^{18}$ wherein R^{18} represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms may be prepared by the reaction of a compound of general formula XI with an acylating agent such as $R^{18}COCl$ in the presence of a catalyst such as aluminium chloride and in an inert organic solvent such as 1,1,2,2-tetrachloroethane and at a temperature from 0°C to the reflux temperature of the reaction mixture.

When Z is an amino group it may also be acylated and subsequent hydrolysis using an acid such as hydrochloric or hydrobromic acid in a solvent such as dioxan or acetic acid may be necessary.

Compounds of general formula I wherein Y represents $-C(=O)R^{18}$ may also be prepared by the reaction of nitriles of the general formula I wherein Y represents $-CN$ with an organometallic reagent such as a compound of general formula $R^{18}MgX^1$ in an inert organic solvent such as diethyl ether or tetrahydrofuran, at a temperature from 0°C to reflux.

Compounds of general formula IC may be prepared by the reaction of a compound of general formula I wherein Y represents the thiocyanato group with an organometallic reagent such as a compound of general formula $RMgX^1$ in an inert organic solvent such as

diethyl ether or tetrahydrofuran, and at a temperature from ambient temperature to the reflux temperature of the reaction mixture.

Compounds of general formula IC wherein RS is other than a 1-alkenylthio group may also be prepared by reacting a compound of general formula I wherein Y represents the thiocyanato group with a base, preferably sodium hydroxide, or a reducing agent, preferably sodium borohydride, in the presence of a reagent of general formula $R'X^1$ wherein R' is as hereinbefore defined for R with the exclusion of 1-alkenyl groups, for example methyl iodide in an inert organic or aqueous-organic solvent such as an alcohol e.g. ethanol or a mixture of an alcohol and water, the reaction being performed at a temperature from ambient to reflux.

Alternatively compounds of general formula IC wherein RS is other than a 1-alkenylthio group may be prepared by reductive alkylation of disulphides of general formula XVII employing a reducing agent preferably sodium dithionite or sodium borohydride, in the presence of a base, preferably sodium hydroxide or sodium carbonate, and of a reagent of general formula $R'X^1$ such as methyl iodide in an inert organic or aqueous-organic solvent such as an alcohol e.g. ethanol or a mixture of an alcohol and water, at a temperature from ambient to reflux.

Alternatively compounds of general formula IC may be prepared from a halide of general formula I wherein Y represents a bromine or iodine atom by metal exchange using a strong base, preferably butyl lithium, and subsequent addition of the appropriate disulphide of general formula R-S-S-R in an inert organic solvent such as tetrahydrofuran, and the reaction is performed at a temperature from -78°C to ambient.

10 Alternatively, compounds of general formula IC wherein RS represents a straight- or branched-chain alkylthio group containing from 1 to 6 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms may be prepared by reacting a compound of general formula XI with an alkanesulphenyl halide (which may be optionally substituted by one or more halogen atoms) in an inert organic solvent, preferably chloroform, in the presence of a base such as pyridine, and at temperatures from 0° to reflux.

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Compounds of general formula IC wherein RS
represents a methylthio group which is substituted by
three halogen atoms which may be the same or different
may also be prepared by the reaction of a compound of
general formula I wherein Y represents the thiocyanato
group with a source of halogenocarbene, such as
chloroform and sodium hydroxide, preferably with phase
transfer catalysis using for example benzyl-
triethylammonium chloride or tetrabutylammonium
chloride.

Compounds of general formula IC wherein RS
represents a straight- or branched-chain alkylthio
group containing from 1 to 6 carbon atoms which is
substituted by one or more fluorine atoms may also be
prepared by a halogen exchange reaction of a compound
of general formula IC wherein RS represents a
straight- or branched-chain alkylthio group containing
from 1 to 6 carbon atoms which is substituted by one
or more chlorine atoms with a fluorinating agent such
as

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as a mixture of antimony trifluoride and antimony pentachloride, KF or CsF in an aprotic solvent such as sulfolane at a temperature from 50°C to reflux.

Compounds of general formula I wherein Y represents the thiocyanato group may be prepared by the reaction of a compound of general formula XI with a thiocyanating agent such as alkali metal or ammonium salts of thiocyanic acid (e.g. NaSCN) and bromine in an inert organic solvent such as methanol, and at a temperature from 0°C to 100°C.

Intermediates of general formula XVII may be prepared by hydrolysis of thiocyanates of general formula I wherein Y represents the thiocyanato group, preferably using hydrochloric acid in the presence of ethanol at a temperature from ambient to reflux temperature; they may also be prepared by reduction of the thiocyanates by sodium borohydride in an alcohol preferably ethanol at a temperature from ambient to reflux.

Compounds of general formula I wherein Y represents a group RSO may be prepared by the oxidation of compounds of general formula IC by an oxidising reagent preferably 3-chloroperbenzoic acid in an inert organic solvent such as dichloromethane or by hydrogen peroxide in acetic acid at a temperature from 0°C to the reflux temperature of the reaction medium.

Compounds of general formula I wherein Y represents a group RSO_2 may also be prepared by the above process, by employing an excess of the oxidising agent.

5 Compounds of general formula I wherein Y represents a group RSO_2 wherein R represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which is substituted by one or more fluorine atoms may also be prepared by a
10 halogen exchange reaction of a compound of general formula I wherein Y represents a group RSO_2 wherein R represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which is substituted by one or more chlorine atoms with a
15 fluorinating agent such as a mixture of antimony trifluoride and antimony pentachloride, KF or CsF at a temperature from $50^\circ C$ to $200^\circ C$.

Compounds of general formula I wherein Y represents a group RSO_2 may also be prepared by
20 reaction of a compound of general formula XI with the appropriate sulphonic anhydride of general formula $(RSO_2)_2O$ for example trifluoromethanesulphonic or
25 methanesulphonic anhydride and in the presence of aluminium chloride as catalyst, and employing an inert organic solvent such as 1,1,2,2-tetrachloroethane at a temperature from ambient to $150^\circ C$.

Compounds of general formula I wherein Z represents a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms, the carboxy group, a group $R^{19}S$ wherein R^{19} represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which may be unsubstituted or substituted by one or more halogen atoms or Z represents a trialkylsilyl group containing from 1 to 6 carbon atoms in each alkyl group which may be the same or different may be prepared by the reaction of a compound of general formula I wherein Z represents a hydrogen, bromine or iodine atom with a lithiating agent preferably lithium diisopropylamide or n-butyl lithium, and reaction with the appropriate substrate from alkyl halide, carbon dioxide, dialkylsulphides or trialkylsilyl halides respectively at a temperature from -78°C to ambient temperature, and in an inert solvent, preferably tetrahydrofuran.

Compounds of general formula I wherein Z represents a hydrogen atom may be prepared by diazotisation of an amine of general formula I wherein Z represents the unsubstituted amino group using an alkyl nitrite, preferably tert-butyl nitrite, in an inert solvent preferably tetrahydrofuran, at a temperature from ambient temperature to the reflux temperature of the reaction mixture.

Compounds of general formula I wherein Z
represents a group $R^{19}SO$ may be prepared by the
reaction of a compound of general formula I wherein Z
represents a group $R^{19}S$ with an oxidising agent,
5 preferably 3-chloroperbenzoic acid in a solvent such
as dichloromethane, or by hydrogen peroxide in acetic
acid at a temperature from 0°C to the reflux
temperature of the reaction mixture.

Compounds of general formula I wherein Z
10 represents a group $R^{19}SO_2$ may also be prepared by
the above process, by employing an excess of the
oxidising agent.

Compounds of general formula I wherein Z
represents the fluorine atom or the cyano group may be
15 prepared by the reaction of a halide of general
formula I wherein Z represents the chlorine or bromine
atom with an alkali metal fluoride, preferably caesium
fluoride, or with an alkali metal cyanide preferably
KCN, under anhydrous conditions in an inert solvent,
20 preferably sulfolane, and at a temperature from
ambient temperature to 150°C.

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Compounds of general formula I wherein Z represents the nitro group may be prepared by oxidation of amines of general formula I wherein Z represents the unsubstituted amino group with an 5 oxidant, preferably trifluoroperacetic acid or m-chloroperbenzoic acid and in an inert organic solvent preferably dichloromethane at a temperature from 0°C to reflux.

Compounds of general formula I wherein Z 10 represents the cyano group may be prepared by dehydrogenation of the corresponding amide preferably by heating with phosphorous pentoxide at a temperature from 50°C to 250°C.

The amides may be prepared (i) by reacting a 15 carboxylic acid of general formula I wherein Z represents a carboxy group with a chlorinating agent preferably thionyl chloride, and (ii) reacting the resultant acid chloride of general formula XVIII with ammonia:-

(i) the reaction with a chlorinating agent preferably thionyl chloride is generally conducted at a temperature from ambient temperature to the reflux temperature of the reaction mixture;

(ii) the reaction with ammonia is generally conducted 20 in a solvent which may be inert, preferably toluene, or in the presence of water, and at a temperature from 0°C to 100°C.

Compounds of general formula I wherein Y represents a group RSO_2 is other than a 1-alkenylsulphonyl group may be prepared alternatively by reaction of sulphinate metal (e.g. sodium) salts with a reagent of general formula $R'X^1$ or preferably a sulphate of general formula $(R')_2SO_4$, in a solvent such as water and in the presence of sodium bicarbonate at a temperature from 0°C to 100°C.

The intermediate sulphinate sodium salts may be prepared by reaction of sulphonyl chlorides of general formula XIV with sodium sulphite in the presence of sodium bicarbonate and water as solvent, at a temperature from 50°C to reflux.

Intermediates of general formula XIV may also be prepared from the thiocyanates of general formula I wherein Y represents a thiocyanato group by chlorination using chlorine in a solvent, preferably water, at a temperature from ambient to reflux.

Compounds of general formula I wherein R^3 represents a haloalkylsulphanyl group may be prepared by oxidation of a haloalkylthio derivative of general formula I, preferably with m-chloroperbenzoic acid in an inert organic solvent preferably dichloromethane, at a temperature from 0°C to reflux.

Compounds of general formula I wherein R^3 represents a haloalkylsulphonyl group may be prepared in a similar manner, by employing two molar equivalents of oxidant.

Compounds of general formula I wherein Y represents the fluorine atom may be prepared by diazotisation of corresponding amines using sodium nitrite in tetrafluoroboric acid and sulphuric acid at a temperature from -10°C to +10°C, followed by photolysis in the presence of excess sodium tetrafluoroboric acid at a temperature from -30°C to ambient.

Intermediate amines above may be prepared by reduction of nitro compounds of general formula I wherein Y represents a nitro group, preferably with zinc in ethanol at a temperature from ambient to reflux.

Compounds of general formula I wherein Y represents the methyl group may be prepared by reduction of an acid of general formula XVI using a reducing agent, preferably borane-tetrahydrofuran complex in a solvent preferably tetrahydrofuran at a temperature from -30°C to reflux.

Compounds of general formula I wherein Z represents a trialkylsilylmethyl group as hereinbefore defined may be prepared by the reaction of a compound of general formula I wherein Z represents the methyl group with a lithiating agent preferably lithium diisopropylamide or n-butyl lithium, and reaction with a trialkylsilyl halide at a temperature from -78°C to ambient, and in an inert organic solvent preferably tetrahydrofuran, optionally in an inert atmosphere.

The following processes optionally followed by the subsidiary processes hereinbefore described permit the preparation of the remaining compounds of general formula I not described above, as well as some whose preparation is described above.

Compounds of general formula I wherein R⁴ represents a chlorine, bromine or iodine atom and Z represents the unsubstituted amino group, may be prepared by the diazotisation of the (diamino) compounds of general formula I in which Z represents and R⁴ is replaced by amino, using a molar equivalent of sodium nitrite in a mineral acid, for example a mixture of concentrated sulphuric acid and acetic acid, at a temperature from 0 to 60°C, and by subsequent reaction with the appropriate copper salt and appropriate mineral acid or with an aqueous solution of potassium iodide (when R⁴ represents an iodine atom) at a temperature from 0 to 100°C.

The diamino compounds above wherein Y represents the cyano group may be prepared by the reaction of potassium cyanoform KC(CN)₃ with a phenylhydrazine of general formula II in the presence of hydrochloric acid, at a temperature from 50°C to the reflux temperature of the reaction mixture.

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Compounds of general formula I wherein R⁴ represents the fluoromethyl group may be prepared by reacting a compound of general formula XII with a fluorinating agent, preferably diethylaminosulphur trifluoride, in an inert organic solvent, preferably dichloromethane, at a temperature from -78°C to the reflux temperature of the reaction mixture.

Intermediates of general formula XII may be prepared by reduction of compounds of general formula XIX preferably with lithium borohydride in an inert organic solvent e.g. tetrahydofuran at a temperature from 0°C to the reflux temperature of the reaction mixture.

Intermediates of general formulae XIX (wherein R²⁰ represents an alkyl group) and XX wherein Z represents the unsubstituted amino group may be obtained by the reaction of a compound of the general formula XIII (wherein R⁰ represents an alkoxy group) with a molar equivalent of compound of general formula Y'CH₂CN, i.e. malononitrile when Y represents the cyano group, in the presence of an anhydrous solvent, e.g. ethanol, and a molar equivalent of a base, e.g. sodium hydride, and at a temperature from 0 to 50°C followed, if desired, by hydrolysis of the esters of general formula XIX with an aqueous base, e.g. sodium hydroxide, with a co-solvent, e.g. ethanol, at a temperature from 0°C to the reflux temperature of the reaction mixture.

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Intermediates of general formulae IV and XIII may be prepared by chlorination of the appropriate unsubstituted compound using chlorine or other chlorinating agent.

5 Intermediates of general formulae IV and XIII may be prepared by diazotisation of the appropriate aniline with a solution of a molar equivalent of sodium nitrite in a mineral acid, e.g. a mixture of concentrated sulphuric acid and acetic acid at a
10 temperature from 0 to 60°C, and then reacting with the compound of formula $\text{CH}_3\text{COCH}(\text{Cl})\text{CN}$ or a compound of general formula $\text{CH}_3\text{CCCH}(\text{Cl})\text{COR}^0$ wherein R^0 represents an alkoxy group in the presence of an inert solvent, e.g. a mixture of water and ethanol,
15 buffered, e.g. with excess sodium acetate, and at a temperature from 0 to 50°C.

Compounds of general formula I wherein R^4 represents the nitro group may be prepared by oxidation of the corresponding amine with an oxidant, 20 preferably trifluoroperacetic acid or *m*-chloroperbenzoic acid in an inert organic solvent preferably dichloromethane at a temperature from 0°C to reflux. By employing known protecting agents in this process compounds of general formula I wherein Z represents the amino group may be prepared.

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Compounds of general formula I wherein R^4 represents the fluorine atom may be prepared by the diazotisation of the corresponding amine of general formula I in which R^4 is replaced by $-NH_2$ using for example a solution of sodium nitrite in a mineral acid, for example sulphuric acid and in the presence of fluoroboric acid or its sodium salt and subsequent thermolysis or photolysis of the diazonium fluoroborate derivative by methods known per se.

Amine intermediates above wherein Z represents the hydrogen atom may be prepared by performing a Curtius rearrangement of the corresponding acid azide by heating in an inert organic solvent such as toluene at a temperature from 50° to 150°C to give an isocyanate which is then reacted with for example tert-butanol to give a carbamate, which in turn is hydrolysed using dilute acid preferably hydrochloric acid in ethanol at a temperature from ambient to reflux.

Intermediate acid azides may be prepared by reaction of a carboxylic acid of general formula XX wherein Z represents the hydrogen atom with a chlorinating agent, preferably thionyl chloride at temperatures from ambient to reflux, followed by reaction of the intermediate acid chloride with sodium azide in a polar solvent, preferably acetone and water at a temperature from 0°C to ambient.

Compounds of general formula I wherein R⁴ represents the cyano group may also be prepared by reacting a carboxylic acid of general formula XX with a chlorinating agent, preferably thionyl chloride at ambient to reflux temperature, followed by reaction of the intermediate acid chloride with ammonia to give an intermediate amide which is then dehydrated by heating with a dehydrating reagent, preferably phosphorus pentoxide at a temperature from 50-250°C.

Intermediates of general formula XX may be prepared by hydrolysis of the corresponding esters of general formula XIX preferably using a base such as sodium hydroxide and a solvent such as aqueous alcohol, and at a temperature from 0°C to the reflux temperature of the reaction mixture.

Compounds of general formula I wherein Y represents a 1,1-difluoroalkyl group which may be substituted by one or more additional halogen atoms may be prepared by the reaction of a compound of general formula I wherein Y represents a straight- or branched-chain alkanoyl group containing from 2 to 6 carbon atoms or the corresponding compound in which Y is replaced by the formyl group or a straight- or branched-chain alkanoyl group containing from 2 to 6 carbon atoms which is substituted by one or more halogen atoms with a fluorinating agent, preferably

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diethylaminosulphur trifluoride or sulphur
tetrafluoride in an inert organic solvent, preferably
dichloromethane, at a temperature from -78°C to
ambient.

5 Compounds of general formula I wherein Y
represents the trifluoromethyl group or a
trifluoromethylalkyl group containing from 2 to 6
carbon atoms which may be substituted by one or more
additional halogen atoms may be prepared by the
10 reaction of a fluorinating agent, e.g. sulphur
tetrafluoride, with an acid of general formula XVI or
the corresponding carboxyalkyl compound (it being
understood that the carboxy group may be attached to
any position of the alkyl moiety) at a temperature
15 from ambient to 150°C.

Salts with pesticidally-acceptable bases of
compounds of general formula I wherein Z represents
the carboxy group may be prepared from the
corresponding compounds of general formula I by
20 methods known per se, for example by reacting
stoichiometric quantities of the compound of general
formula I and the appropriate base, for example an
alkali metal hydroxide, carbonate or bicarbonate, an
alkaline earth metal hydroxide or carbonate, ammonia
25 or an amine (e.g. diethanolamine, triethanolamine,
octylamine, morpholine or dioctylamine), in a suitable
solvent. The salts may, if necessary, be purified by
recrystallisation from one, two or more suitable
solvents.

Compounds of general formula I not hitherto disclosed or described in the chemical literature, together with their processes of preparation form further features of the present invention.

5 The present invention accordingly provides the compounds of general formula I, wherein the various symbols are as hereinbefore defined, and salts thereof, with the exclusion of the compounds wherein: R⁴ and Z both represent methyl, Y represents thiocyanato and (R³)_n represents 2-, 3- or 4-nitro, 4-methyl, 4-chloro or 2,4-dinitro substitution; R⁴ represents methyl, Y represents cyano, Z represents unsubstituted amino and (R³)_n represents 4-chloro, 2,4-dichloro, 3,4-dichloro, 15 3-chloro-4-methyl or 2-methyl-4-chloro substitution; R⁴ represents methyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n represents 3- or 4-fluoro substitution; R⁴ represents ethyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n 20 represents 3- or 4-chloro, 2-, 3- or 4-fluoro or methyl, 3-bromo or 3-nitro substitution; R⁴ represents propyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n represents 3-fluoro substitution; R⁴ represents methyl, Y represents sulphamoyl, Z represents chloro and (R³)_n represents 4-chloro substitution; R⁴ 25

represents methyl, Y represents nitro, and Z represents chloro or R⁴ represents chloro, Y represents nitro, and Z represents methyl and (R³)_n represents 4-nitro; and R⁴ represents 5 nitro, Y represents cyano or CONH₂, Z represents hydrogen and (R³)_n represents 4-nitro substitution.

According to a further feature of the present invention there are provided intermediates for the preparation of certain compounds of general formula I i.e. compounds for which is their alternative meanings: 10 Y represents the hydrogen atom, the formyl or carboxy group, a straight- or branched-chain alkanoyl group containing from 2 to 6 carbon atoms which is substituted by one or more halogen atoms, the dithio group (which joins two pyrazole rings), the amino group, the -SO₂Cl group, a straight- or 15 branched-chain carboxyalkyl group containing from 2 to 6 carbon atoms, Z represents the carbamoyl group or a straight- or branched-chain alkoxy carbonyl group containing from 2 to 7 carbon atoms or the diphenoxycarbonylamino group, (R³)_n substitution is a preferred combination given earlier in the specification or R⁴ represents the amino, 20 hydroxymethyl, carboxy or carbamoyl group or a straight- or branched-chain alkoxy carbonyl or 25 alkoxy carbonylamino group containing from 2 to 7 carbon atoms.

The following Examples and Reference Examples illustrate the preparation of compounds of general formula I according to the present invention:-

EXAMPLE 1

5 Compound No.1

A mixture of 2,4,6-trichlorophenylhydrazine (21.1 g) and tetracyanoethylene (13.3 g) in ethanol (100 ml) was heated at reflux for 15 minutes. The reaction mixture was cooled and the solid precipitate was filtered off and washed with diethyl ether to give 10 5-amino-3,4-dicyano-1-(2,4,6-trichlorophenyl)pyrazole (13 g), as a buff coloured solid, m.p. 267-271°C.

EXAMPLE 2

15 Compounds Nos.2 and 3

Tetracyanoethylene (1.9 g) and 2,6-dichloro-16 4-trifluoromethylphenylhydrazine (3.7 g) was added to a magnetically-stirred solution of sodium acetate (0.6 g) in glacial acetic acid (15 ml) at laboratory 21 temperature. After stirring for 15 minutes, a colourless solid precipitated from the solution and stirring was continued overnight. The mixture was 26 then filtered. The solid obtained was washed successively with acetic acid, water, aqueous sodium bicarbonate solution and water, to give 25 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole (2.5 g), as beige crystals, m.p. 221-222°C.

By proceeding in a similar manner, but replacing the 2,6-dichloro-4-trifluoromethylphenylhydrazine by 2,3,5,6-tetrachlorophenylhydrazine, there was obtained:-

5 5-Amino-3,4-dicyano-1-(2,3,5,6-tetrachlorophenyl)pyrazole, m.p. greater than 330°C, in the form of a buff-coloured powder.

REFERENCE EXAMPLE 1

Phenylhydrazines used as starting materials in Examples 1, 2 and 11, not hitherto described in the chemical literature were prepared as follows:-

2,6-Dichloro-4-trifluoromethylphenylaniline (4.3 g) was dissolved with stirring, in glacial acetic acid (23 ml). A solution of sodium nitrite (1.5 g) in concentrated sulphuric acid (11 ml) was then added at 15 55-60°C. The solution thus obtained was cooled to 0-5°C and a solution of stannous chloride (16.4 g) in concentrated hydrochloric acid (14 ml) was added with vigorous stirring. A cream-coloured solid precipitated. The mixture was filtered and the solid obtained was added to a mixture of aqueous ammonia solution and ice. The mixture thus obtained was extracted with diethyl ether (6 x 500 ml) and the combined ethereal extracts were dried over sodium sulphate, filtered and evaporated to dryness to give 2,6-dichloro-4-trifluoromethylphenylhydrazine (3.7 g) 25 m.p. 54-56°C, in the form of a colourless crystalline solid.

By proceeding in a similar manner, but replacing the 2,6-dichloro-4-trifluoromethylaniline by the hereinafter indicated aniline there were prepared:-

2-Chloro-4-trifluoromethylphenylhydrazine, m.p. 38-39°C, in the form of a colourless solid, from 2-chloro-4-trifluoromethylaniline.

EXAMPLE 3

Compound No.4

Ethoxyethylenemalononitrile (44.5g) and 2,6-dichloro-4-trifluoromethylphenylhydrazine (80.0g) were added to a stirred solution of sodium acetate (13.4g) in glacial acetic acid (110 ml) at laboratory temperature. A thick suspension was obtained and was stirred overnight, after which a dark solution had formed. The solvent was evaporated in vacuo, and the residue was diluted with aqueous sodium bicarbonate solution (100ml) and extracted with dichloromethane (3 x 100ml), and the combined extracts were washed with sodium bicarbonate solution (50ml), then with water (100ml), dried over anhydrous magnesium sulphate and evaporated in vacuo to give a dark syrup. This was heated at reflux with 2-ethoxyethanol (200ml) for 1 hour, and then evaporated in vacuo to give a dark oil. The oil was dissolved in dichloromethane, washed with sodium bicarbonate solution (50ml), then with water (100ml), dried over anhydrous magnesium sulphate, treated with charcoal, and evaporated in vacuo to give

a black solid. The solid was recrystallised twice from a mixture of toluene and petroleum ether (b.p. 60-80°C) to give 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole (49.3g), m.p. 194-196°C, in the form of pale brown crystals.

EXAMPLE 4

Compounds Nos. 5, 22, 24 and 36

To a mechanically stirred solution of 2,6-dichloro-4-trifluoromethylphenylhydrazine (180.3 g) in dry diethyl ether (700 ml) was added anhydrous potassium carbonate (112 g), and the mixture was cooled to 0°C. To this mixture was added dropwise during half-an-hour a solution of 2-chloro-1,1-dicyano-2-trifluoromethylethylene (132.1 g) in dry diethyl ether (350 ml). The ice-bath was removed at the end of the reaction, and the mixture was left overnight and then poured onto water (2000 ml). The ethereal layer was separated and the aqueous solution extracted with diethyl ether (2 x 300 ml). The combined extracts were dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo to give a buff solid (350 g). Recrystallisation from toluene/hexane gave white crystals (169.5 g) m.p. 202-204°C of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole.

By proceeding in a similar manner, but
replacing the 2-chloro-1,1-dicyano-2-trifluoromethyl-
ethylene by 2-chloro-1-cyano-1-methanesulphonyl-
2-trifluoromethylethylene there was prepared:-

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole
in the form of buff crystals m.p. 215-218°C, from
toluene-hexane.

By proceeding in a similar manner, but
replacing the 2-chloro-1,1-dicyano-2-trifluoromethyl-
ethylene by 2-chloro-1-cyano-1-methoxycarbonyl-
2-trifluoromethylethylene there was prepared:-

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-4-methoxycarbonyl-3-trifluoromethylpyrazole in
the form of fawn crystals, m.p. 114-115°C, from hexane.

By proceeding in a similar manner, but
replacing the 2,6-dichloro-4-trifluoromethylphenyl-
hydrazine by 2,6-dichloro-4-trifluoromethoxyphenyl-
hydrazine there was prepared:-

5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoro-
methoxyphenyl)-3-trifluoromethylpyrazole in the form
of white crystals, m.p. 150-160.5°C from toluene-
hexane.

EXAMPLE 5

Compounds Nos. 19, 20, 21 and 47

Anhydrous sodium acetate (0.246 g) was added to a stirred solution of 2-chloro-1,1-dicyano-2-pentafluorooethyl-ethylene (1.38 g) in acetic acid (5 ml). To this mixture was added 2,6-dichloro-4-trifluoromethylphenylhydrazine (1.47 g) during 5 minutes. After stirring overnight the mixture was neutralised with sodium bicarbonate solution, and extracted with dichloromethane (2 x 50 ml). The combined extracts were washed with water, dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo to give a buff solid (2.1 g). This solid was heated under reflux with 2-ethoxyethanol (10 ml) for 1 hour, and evaporated in vacuo to give a brown oil (2.2 g). This oil was chromatographed on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) using a mixture of dichloromethane and ethyl acetate (98:2) to give a yellow solid. Recrystallisation from a mixture of dichloromethane and petroleum ether gave 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-pentafluoroethylpyrazole as white crystals, m.p. 160-162°C.

By proceeding in a similar manner, but replacing the 2-chloro-1,1-dicyano-2-pentafluoroethyl-ethylene by 2-chloro-1,1-dicyano-2-chlorodifluoromethyl-ethylene there was prepared:-

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5-Amino-3-chlorodifluoromethyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyanopyrazole in the form of white prisms, m.p. 192°C, from toluene-hexane.

5 By proceeding in a similar manner, but replacing the 2-chloro-1,1-dicyano-2-pentafluoroethyl-ethylene by 2-chloro-1,1-dicyano-2-difluoromethyl-ethylene there was prepared:-

10 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-3-difluoromethylpyrazole in the form of a colourless solid, m.p. 184.5°C (from toluene-petroleum ether).

15 By proceeding in a similar manner, but replacing the 2-chloro-1,1-dicyano-2-pentafluoroethyl-ethylene by 2-chloro-1,1-dicyano-2-heptafluoropropyl-ethylene there was prepared:-

20 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-3-heptafluoropropylpyrazole in the form of colourless prisms, m.p. 139-140°C (from toluene-petroleum ether).

REFERENCE EXAMPLE 2

Chloro-dicyanoethylenes used as starting materials in the above Examples, not hitherto described in the chemical literature were prepared as follows:-

A suspension of 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt (18.56 g) in dichloromethane (60 ml) was stirred at room temperature and treated with phosphorus pentachloride (19.27 g). The suspension was heated under reflux for 6 hours, cooled and filtered, and the filtrate was distilled. A Widmer fractionating column was used to give 2-chloro-1,1-dicyano-2-difluoromethyl-ethylene as a liquid, b.p. 88°C (44 mm Hg) (71 g).

10 By proceeding in a similar manner, but replacing 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt by 2-cyano-3-hydroxy-4,4-difluorobut-2-enenitrile sodium salt there was prepared 2-chloro-1,1-dicyano-2-difluoromethyl-ethylene as a liquid, b.p. 94°C (46 mm Hg).

15 By replacing 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt by 3-hydroxy-2-methanesulphonyl-4,4,4-trifluorobut-2-enenitrile sodium salt and proceeding in a similar manner there was prepared 2-chloro-1-cyano-1-methanesulphonyl-2-trifluoromethyl-ethylene as a pale brown liquid.

20 By replacing 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt by 3-hydroxy-2-methoxycarbonyl-4,4,4-trifluorobut-2-enenitrile sodium salt and proceeding in a similar manner there was prepared 2-chloro-1-cyano-1-methoxycarbonyl-2-trifluoromethyl-ethylene as a colourless oil, b.p. 86-92°C at 23-25 mm Hg.

By replacing 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt by 2-cyano-3-hydroxy-4,4,5,5,6,6,6-heptafluorohex-2-enenitrile sodium salt and proceeding in a similar manner there 5 was prepared 2-chloro-1,1-dicyano-2-heptafluoropropyl-ethylene as a pale yellow liquid, b.p. 110°C at 60 mmHg.

REFERENCE EXAMPLE 3

10 The sodium salts used in the above Reference Examples as starting materials, not hitherto described in the chemical literature were prepared as follows:-

15 To a solution of sodium methoxide (5.61 g) in anhydrous methanol (70 ml) was added malononitrile (6.85 g) and the yellow solution treated with methyl chlorodifluoroacetate (15 g). The mixture was heated under reflux for 4 hours, the solvent was evaporated in vacuo and re-evaporated after addition of toluene to give 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt as a brown solid 20 (18.9g). This was dried in a vacuum desiccator.

25 By proceeding in a similar manner, but replacing methyl chlorodifluoroacetate by ethyl difluoroacetate there was obtained 2-cyano-3-hydroxy-4,4-difluorobut-2-enenitrile sodium salt as a light brown solid.

By proceeding in a similar manner, but replacing methyl chlorodifluoroacetate by methyl

trifluoroacetate, and the nitrile by methanesulphonylacetone. There was obtained 3-hydroxy-2-methanesulphonyl-4,4,4-trifluorobut-2-enenitrile sodium salt as a brown solid.

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By proceeding in a similar manner, but replacing methyl chlorodifluoroacetate by methyl-trifluoroacetate, and the malononitrile by methyl-cyanoacetate there was obtained 3-hydroxy-2-methoxy-carbonyl-4,4,4-trifluorobut-2-enenitrile sodium salt as a buff solid.

•10

By proceeding in a similar manner, but replacing methyl chlorodifluoroacetate by methyl-heptafluorobutyrate there was obtained 2-cyano-3-hydroxy-4,4,5,5,6,6,6-heptafluorohex-2-enenitrile sodium salt as a light brown hygroscopic solid.

EXAMPLE 6

Compound No. 23

To stirred 80% sulphuric acid (22 ml) was added 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.98 g) at 20 80°C. After 1 hour the cooled solution was poured onto ice and extracted with dichloromethane (3x). The combined extracts were washed with water, dried over anhydrous magnesium sulphate, filtered, and 25 evaporated in vacuo to give a white solid. This solid was recrystallised from ethyl acetate-petroleum ether to give 5-amino-4-carbamoyl-1-(2,6-dichloro-

4-trifluoromethylphenyl)-3-trifluoromethylpyrazole
(3.5 g), m.p. 169-171°C in the form of white crystals.

EXAMPLE 7

Compounds Nos. 6, 7 and 8

3,5-Diamino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole (3.9g; prepared as described below) was dissolved with stirring in glacial acetic acid (60ml) at 15°C. A solution of sodium nitrite (0.88g) in concentrated sulphuric acid (5.85ml) was then added over 5 minutes, maintaining at 15°C. After 15 minutes longer at this temperature, the dark red oil solution was poured during 1 minute onto a stirred solution of cuprous chloride (2.32g) in concentrated hydrochloric acid (36ml). After 15 minutes at laboratory temperature, by which time the evolution of nitrogen had completely subsided, the reaction mixture was poured onto excess ice and water, and extracted with dichloromethane (3 x 50ml). The combined extracts were washed with water (2 x 50ml), then with sodium bicarbonate solution (50ml), dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a brown semi-solid (4.1g).

Chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) using a mixture of dichloromethane and ethyl acetate (98:2) as eluent gave after evaporation of the eluate and recrystallisation of the residue from a mixture of dichloromethane and petroleum ether

(b.p. 60-80°C) 5-amino-3-chloro-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole (0.95g), m.p. 189-191°C, in the form of white crystals.

By proceeding in a similar manner but
5 replacing the cuprous chloride and concentrated
hydrochloric acid by cuprous bromide and 48% w/v
- hydrobromic acid respectively there was prepared:-

10 5-Amino-3-bromo-4-cyano-1-(2,6-dichloro-4-
trifluoromethylphenyl)pyrazole, m.p. 182-183°C, in the
form of white crystals.

By replacing the cuprous chloride and
concentrated hydrochloric acid by a solution of
potassium iodide in water there was prepared:-

15 5-Amino-3-iodo-4-cyano-1-(2,6-dichloro-4-
trifluoromethylphenyl)pyrazole, m.p. 208-210°C, in the
form of white crystals.

REFERENCE EXAMPLE 4

A suspension of 2,6-dichloro-4-trifluoro-
methylphenylhydrazine (14.7g) in water (40ml) was
20 stirred with concentrated hydrochloric acid (5.2ml),
and potassium cyanoform (8.52g) added. The suspension
was stirred and heated under reflux for 16 hours, and
left to cool overnight. The mixture was washed into a
separating funnel with the aid of ethyl acetate and
25 water, and the organic phase collected. The aqueous
phase was re-extracted with ethyl acetate (2 x 80ml),
and the combined organic solutions washed with water

(2 x 50ml), dried over anhydrous magnesium sulphate, and evaporated in vacuo to give an orange solid (20.9g). Two recrystallisations from a mixture of ethyl acetate and petroleum ether (b.p. 60-80°C) gave 5 3,5-diamino-4-cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)pyrazole (7.75g), m.p. 208-210°C in the form of white crystals.

EXAMPLE 8

Compound No.9

10 A solution of ethanethiol (2.1g) in toluene (10ml) was added dropwise at 5-10°C to a stirred suspension of N-chlorosuccinimide (4.7g) in toluene (40ml). The reaction mixture was filtered after 20 minutes to give a solution of ethanesulphenyl chloride. This filtrate was added dropwise with stirring to a solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole sodium salt [prepared in situ by reaction of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole (5g) with sodium hydride (0.4g)] in tetrahydrofuran (50ml) containing 15-crown-5 (3 drops) at a temperature of 5-10°C. After 2 hours, aqueous sodium bicarbonate solution (50ml) was added, and the organic phase was separated and washed with water (2 x 50ml), and dried over anhydrous magnesium sulphate. Evaporation of the solvent in vacuo gave a dark brown gum, which was chromatographed on silica (Merck

- 10c.

230-400 mesh, 0.7 kg cm⁻²) using dichloromethane as eluent. Evaporation of the eluates gave an orange gum, which then recrystallised from a mixture of ethyl acetate and hexane to give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-5-ethanesulphenylamino-pyrazole (2.3g), m.p. 160-161°C, in the form of a pale yellow solid.

EXAMPLE 9

Compounds Nos. 10, 11 and 27

10 A mixture of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole (5g) and p-toluenesulphoic acid hydrate (0.1g) in trimethylorthoformate (20ml) was heated at reflux for 4.5 hours. After cooling, the reaction mixture was evaporated to dryness in vacuo. The residue was dissolved in diethyl ether and left to crystallise at 0°C. The dark coloured solid was recrystallised from a mixture of ethanol and water to give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-5-methoxymethyleneaminopyrazole (4.67g), m.p. 75-78°C, in the form of buff crystals.

By proceeding in a similar manner but replacing the trimethylorthoformate by tripropyl-orthoformate there was prepared:-

25 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-5-propoxymethyleneaminopyrazole, m.p. 77-79°C, in the form of buff crystals.

230-400 mesh, 0.7 kg cm⁻²) using dichloromethane as eluent. Evaporation of the eluates gave an orange gum, which then recrystallised from a mixture of ethyl acetate and hexane to give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-5-ethanesulphenylamino-pyrazole (2.3g), m.p. 160-161°C, in the form of a pale yellow solid.

EXAMPLE 9

Compounds Nos. 10, 11 and 27

10 A mixture of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole (5g) and p-toluenesulphonic acid hydrate (0.1g) in trimethylorthoformate (20ml) was heated at reflux for 4.5 hours. After cooling, the reaction mixture was evaporated to dryness in vacuo. The residue was dissolved in diethyl ether and left to crystallise at 0°C. The dark coloured solid was recrystallised from a mixture of ethanol and water to give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-5-methoxymethyleneaminopyrazole (4.67g), m.p. 75-78°C, in the form of buff crystals.

By proceeding in a similar manner but replacing the trimethylorthoformate by tripropyl-orthoformate there was prepared:-

25 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-5-propoxymethyleneaminopyrazole, m.p. 77-79°C, in the form of buff crystals.

buff-coloured solid (23.0g). Recrystallisation from a mixture of ethyl acetate and petroleum ether (b.p. 60-80°C) gave 5-acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, m.p. 208-209°C, in the form of white crystals.

By proceeding in a similar manner, the following phenylpyrazoles were obtained by acylation of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole with the appropriate acid chloride:-

5-Dichloracetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, m.p. 186-187°C after purification by trituration with carbon tetrachloride and subsequent recrystallisation from a mixture of ethanol and water, in the form of an off-white solid. The reaction was performed at laboratory temperature.

5-Cyclopropylcarbonamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, m.p. 217-218°C after recrystallisation from a mixture of ethanol and water, in the form of an off-white solid. The reaction was performed at laboratory temperature.

5-Pentanamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole in the form of a pale yellow glass. Infra-Red Absorption bands: 3260, 3100, 2960, 2940, 2880, 2240, 1730, 1700, 1315, 880, 820 cm^{-1} (liquid film). The reaction was

performed at 0°C during the addition, and at laboratory temperature thereafter.

5 5-Propionamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, m.p. 188-189°C after purification by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) using a mixture of acetone and hexane (2:3) as eluent, and subsequent trituration with toluene, in the form of a white powder. The reaction was performed at laboratory 10 temperature.

15 By proceeding in a similar manner, but replacing the solvent by acetonitrile, the following phenylpyrazole was obtained by acylation of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyano-20 pyrazole with trimethylacetyl chloride:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-(2,2-dimethylpropionamido)pyrazole as white crystals, m.p. 202-203°C from toluene-hexane, and after purification by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) using a mixture of dichloromethane and ethyl acetate (9:1) as eluent.

25 By proceeding in a similar manner but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole by 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole and by heating under reflux for 18 hours there was obtained:-

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5-Acetamido-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole, m.p. 225-227°C, from ethyl acetate-hexane, in the form of white crystals.

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EXAMPLE 11Compounds Nos. 17 and 18

Anhydrous sodium acetate (1.0g) was dissolved in stirred acetic acid (40ml), and tetracyanoethylene (3.5g) was added at laboratory temperature. 2-Chloro-4-trifluoromethylphenylhydrazine (5.25g) was added in one portion, and the mixture was stirred overnight. After dilution with water, the precipitated solid was filtered off to give, after drying, 5-amino-1-(2-chloro-4-trifluoromethylphenyl)-3,4-dicyano pyrazole, m.p. 209-210°C in the form of a white powder.

By proceeding in a similar manner but replacing the 2-chloro-4-trifluoromethylphenylhydrazine by 2,3,5,6-tetrafluoro-4-trifluoromethylphenylhydrazine and with cooling during addition of the phenylhydrazine to the tetracyanoethylene solution, there was prepared:-

5-amino-3,4-dicyano-1-(2,3,5,6-tetrafluoro-4-trifluoromethylphenyl)pyrazole, m.p. 252-263°C in the form of a buff powder.

EXAMPLE 12

Compounds Nos. 28 and 29

Sodium hydride (80%, 0.25 g) was added to a stirred solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (2.9 g) in dry tetrahydrofuran (50 ml). After 3 hours at room temperature, 15-crown-5 (1 drop) and methyl iodide (2 g) was added at 0°C, and the mixture left overnight at room temperature. The solution was evaporated in vacuo, and the residue was dissolved in dichloromethane (50 ml), washed with water, dilute hydrochloric acid and water. After drying over anhydrous magnesium sulphate, filtration, and evaporation in vacuo a yellow oil was obtained.

Purification by chromatography using Merck silica (230-400 mesh, 0.7 kg cm⁻²) with dichloromethane as eluent gave 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-dimethylamino-3-trifluoromethylpyrazole as a white solid, m.p. 105-107°C.

By proceeding in a similar manner but replacing the methyl iodide by ethyl bromoacetate, and employing dioxan as solvent in place of tetrahydrofuran there was obtained 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-ethoxycarbonylmethylamino-3-trifluoromethylpyrazole as white crystals, m.p. 104-106°C from ethyl acetate-petroleum ether.

EXAMPLE 13Compound No. 30

To a suspension of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-ethoxymethyleneamino-3-trifluoromethylpyrazole (1.0 g) in methanol (10 ml) stirred at room temperature was added sodium borohydride (0.17 g). After 2 hours an additional 0.17 g of sodium borohydride was added, and another 0.34 g added after 1 hour. One hour later the mixture was poured onto water (80 ml) and extracted with dichloromethane (3 x 25 ml). The combined extracts were dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo. The white solid thus obtained was purified by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) with dichloromethane as eluent, to furnish 4-cyano-5-methylamino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole as a white solid (0.6 g), m.p. 200-202°C.

EXAMPLE 14Compounds Nos. 31, 37 and 38

Sodium hydride (80%, 0.3 g) was added to a stirred solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (2.9 g) in dry tetrahydrofuran (50 ml). After 3 hours, 15-crown-5 (1 drop) and trimethylacetyl chloride (1.8 g) was added, and the mixture stirred

overnight. Evaporation in vacuo gave a buff semisolid, which was dissolved in dichloromethane. This solution was washed with water, dilute hydrochloric acid and with water again and finally dried over anhydrous magnesium sulphate. Filtration followed by evaporation in vacuo gave a yellow oil, which was purified by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm⁻²). Elution with dichloromethane gave after evaporation 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-(2,2-dimethylpropionamido)-3-trifluoromethylpyrazole as a white solid, m.p. 198-200°C.

By proceeding in a similar manner but replacing the trimethylacetyl chloride by ethyl chloroformate there was obtained, after recrystallisation from toluene, 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-3-trifluoromethylpyrazole as white crystals, m.p. 62°C.

By proceeding in a similar manner but replacing the trimethylacetyl chloride by cyclopropanecarboxylic acid chloride there was obtained 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(cyclopropanecarbonyl)amino-3-trifluoromethylpyrazole as a pale yellow solid, m.p. 126-127°C.

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EXAMPLE 15Compound No.39

A solution of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(cyclopropanecarbonyl)-amino-3-trifluoromethylpyrazole (1.0 g) in ethanol (50 ml) was heated under reflux with saturated sodium bicarbonate solution (25 ml) for 45 minutes. After cooling, and evaporation of the solvent in vacuo, the residue was diluted with water and extracted with dichloromethane. The extract was dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo to give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-cyclopropanecarbonamido-3-trifluoromethylpyrazole as a white solid m.p. 210-212°C.

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EXAMPLE 16Compound No.33

A stirred mixture of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.89 g) and bromoform (13 ml) was treated with tert-butyl nitrite (2.26 ml) at room temperature. After 15 minutes the mixture was heated to 50°C for 1 hour, and evaporated in vacuo to yield a red oil. This was purified by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm⁻²) eluting with a mixture of dichloromethane and petroleum ether (1:2) to furnish 5-bromo-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole as a fawn solid m.p. 85-87°C (3.7 g).

EXAMPLE 17

Compound No.34

A solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-hydroxymethylpyrazole (1.25 g) in dichloromethane (10 ml) was added slowly to a stirred solution of diethylaminosulphur trifluoride (0.66 g) in dichloromethane (6 ml) cooled to -78°C. After 30 minutes at this temperature the solution was warmed to room temperature and stirred for 2 hours. The mixture was then poured onto water (20 ml) and the dichloromethane layer was separated, dried over anhydrous magnesium sulphate, filtered and evaporated in vacuo. The product was purified by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm⁻²) eluting with a mixture of dichloromethane and ethyl acetate (98:2), and subsequent recrystallisation from dichloromethane-petroleum ether to give 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-fluoromethylpyrazole as a white solid m.p. 139-141°C.

REFERENCE EXAMPLE 5

5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-hydroxymethylpyrazole was prepared as follows:-

A solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonylpyrazole (1.0 g) in dry tetrahydrofuran (15 ml) was

reated under nitrogen with lithium borohydride (0.06 g) with stirring at room temperature for 18 hours. Ethyl acetate (5 ml) followed by saturated sodium chloride solution (5 ml) was added, and the mixture was acidified with dilute hydrochloric acid and extracted with dichloromethane. The extract was dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo. The residual oil was purified by chromatography on silica (Merck, 40-230 mesh, 10 0.7 kg cm⁻²) eluting with a mixture of dichloromethane and ethyl acetate (1:1), and the pure fractions were evaporated in vacuo and recrystallised from ethyl acetate-petroleum ether to give the title compound as a white solid m.p. 159-161°C

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5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonylpyrazole was prepared as follows:-

To sodium hydride (80%, 0.9 g) in dry ethanol (30 ml) was added, with stirring, malononitrile (1.98 g). Ethyl chloro-(2,6-dichloro-4-trifluoromethylphenyl)hydrazone-acetate (11.0 g) was then added with stirring and cooling. The internal temperature quickly rose to 20°C and was kept at that for 1 hour, before filtration of the pale yellow solid. The filtrate was evaporated in vacuo to give an orange solid. The combined solids were dissolved in ethyl

acetate, washed twice with water, dried over anhydrous magnesium sulphate, filtered and evaporated in vacuo to give an orange solid (11.0 g). Recrystallisation from ethyl acetate-petroleum ether gave the title compound as fawn crystals (8.3 g) m.p. 208-209°C.

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Ethyl chloro-(2,6-dichloro-4-trifluoromethyl-phenyl)hydrazoneacetate was prepared as follows:-

Sodium nitrite (3.04 g) was added during 15 minutes to stirred concentrated sulphuric acid (24 ml) at 30-50°C. The solution was cooled to 20°C, and added dropwise during 15 minutes to a solution of 2,6-dichloro-4-trifluoromethylaniline (9.2 g) in acetic acid (90 ml), maintaining at 35-40°C. This solution was then cooled to +10°, and added dropwise to a stirred solution of anhydrous sodium acetate (54 g) and ethyl chloroacetacetate (7.0 g) in a mixture of water (72 ml) and ethanol (48 ml) during 45 minutes with cooling such that the temperature was kept at 10°C. After 1 hour at room temperature the mixture was diluted with water, filtered, and the solid dissolved in dichloromethane. This solution was dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo to give the title compound as a white solid (11.9 g) m.p. 96-98°C.

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EXAMPLE 18Compounds Nos. 32 and 40

A mixture of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.64 g) and N-bromosuccinimide (1.78 g) in carbon tetrachloride (30 ml) was stirred and heated under reflux for 1 hour. Further N-bromosuccinimide (0.89 g) was added, and reflux was continued for a further 1 hour. The mixture was cooled, filtered, and the filtrate was evaporated in vacuo to give an orange solid. Recrystallisation from petroleum ether gave

15 5-amino-4-bromo-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole in the form of white crystals (2.6 g) m.p. 119-120°C.

By proceeding in a similar manner but replacing N-bromosuccinimide by N-chlorosuccinimide there was obtained 5-amino-4-chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole as white crystals m.p. 99-100°C. No excess of chlorinating agent was required in this case.

REFERENCE EXAMPLE 6

25 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole was prepared as follows:-

A solution of 5-amino-4-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.64 g) in

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methyldiazole (10.5 g) in N,N-dimethylaniline (13 ml) was heated under reflux for 3 hours. The cooled mixture was poured onto concentrated hydrochloric acid (15 ml) and extracted with ether (4 x 30 ml). The combined extract was washed with 6N hydrochloric acid (3 x 30 ml), with water (2 x 30 ml), dried over anhydrous magnesium sulphate, filtrated, and evaporated in vacuo. The product was recrystallised from cyclohexane to give the title compound (5.7 g) as white needles m.p. 126-128°C.

5-Amino-4-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole was prepared as follows:-

A mixture of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methoxycarbonyl-3-trifluoromethylpyrazole (101.2 g; hereinbefore described in Example 4) and sodium hydroxide (43 g) in water (170 ml) and methanol (550 ml) was stirred at room temperature for 2 days, evaporated in vacuo, and the residue triturated with dilute hydrochloric acid. The solid was filtrated, dissolved in ethyl acetate, and the resulting solution was washed with sodium chloride solution. After drying over anhydrous magnesium sulphate, filtration, and evaporation in vacuo a semisolid residue was obtained. This residue was triturated with hexane and the solid was recrystallised from toluene-hexane to give the title compound as a cream solid, m.p. 212-215°C.

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EXAMPLE 19Compound No.41

Ethyl chloroformate (1.6 g) was added to a stirred solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.9 g) in pyridine (15 ml). After stirring overnight another addition of ethyl chloroformate (1.0 ml) was made, and the mixture was left for a further 12 hours. The solvent was evaporated in vacuo and the residue was acidified with dilute hydrochloric acid, and extracted with dichloromethane. This extract was washed with water (3x), dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo. Purification by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm^{-2}) eluting with ethyl acetate - petroleum ether (1:1) gave a white solid, which was recrystallised from a mixture of dichloromethane and hexane to furnish white crystals of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-ethoxycarbonylamino-3-trifluoromethylpyrazole, m.p. 177-179°C.

EXAMPLE 20Compound No.35

A solution of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.0 g) in concentrated sulphuric acid (10 ml) at 0°C was treated with fuming nitric acid (9 ml) during 15

minutes, keeping the temperature at 5-15°C. After 30 minutes the mixture was poured onto excess ice, and the precipitated solid was filtered off and dissolved in ethyl acetate. After drying over anhydrous magnesium sulphate, filtration, and evaporation in vacuo a brown oil was obtained. This oil was dissolved in the minimum of ethyl acetate and diluted with hexane. A pale yellow solid crystallised and this was discarded. The filtrate was evaporated in vacuo to give a solid which was recrystallised from toluene-hexane to furnish a yellow solid. One further recrystallisation from the same solvent pair gave 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole as white crystals, m.p. 214-215°C.

EXAMPLE 21

Compound No.42

To a solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (2.33 g) in dry tetrahydrofuran (30 ml) was added with stirring at room temperature, a solution of *tert*-butyl nitrite (1.36 ml) in dry tetrahydrofuran (5 ml) during 2 minutes. The solution was then heated under reflux for 1 hour and cooled, and additional *tert*-butyl nitrite (2.72 ml) was added. The solution was heated under reflux for 30 minutes, and left to cool

overnight. Evaporation in vacuo gave an orange oil, which was purified by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm⁻²) eluting with dichloromethane- hexane (1:1). The product was finally recrystallised from hexane to give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole, m.p. 121-123°C, as white crystals.

EXAMPLE 22

10 Compound No. 43

To a solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole (2.33 g) in chloroform (30 ml) stirred at room temperature, was added iodine (3.0 g) followed by 15 tert-butyl nitrite (1.1 g). After 2 hours the mixture was heated under reflux for 1.5 hours, cooled and filtered, and the filtrate was washed with sodium thiosulphate solution to remove excess iodine. After washing with water, drying over anhydrous magnesium sulphate and evaporation in vacuo, a yellow solid was obtained. This was chromatographed on silica (Merck, 20 40-230 mesh, 0.7 kg cm⁻²) eluting with dichloro-methane-hexane (1:2) to give a yellow oil. Dissolution in hot hexane gave, on cooling, 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-iodo-3-trifluoromethylpyrazole as white crystals, 25 m.p. 86-87°C.

EXAMPLE 23

Compound No.44

To dry diisopropylamine (0.135 g) in dry tetrahydrofuran (4 ml) stirred at -78°C under nitrogen, was added via a syringe, a solution of α -butyl lithium (0.52 ml of a 2.6 M solution in hexane). After warming to room temperature during 1 minute, the solution was re-cooled to -78°C, and added via a syringe to a stirred solution of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-pyrazole (0.5 g) in dry tetrahydrofuran (4 ml) under nitrogen at -78°C. The addition, during 2 minutes, was exothermic and the internal temperature was maintained at -60°C for a further 15 minutes. Methyl iodide (0.1 ml) was added. After 1.5 hours at this temperature the solution was poured onto excess water and extracted with dichloromethane (3x). The combined organic phase was washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a solid. Chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm^{-2}) eluting with dichloromethane-hexane (1:3) gave a white solid (0.2 g). Recrystallisation from hexane furnished 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-methyl-3-trifluoromethylpyrazole as white crystals, m.p. 90-92°C.

EXAMPLE 24

Compound No.45

A mixture of 5-amino-4-chlorosulphonyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (1.2 g) and dimethylamine (17.6 ml of a 40% aqueous solution) was heated on a steam-bath for 1 hour, cooled, and poured onto crushed ice (50 g) to give a brown solid. This solid was filtered, dried, and recrystallised from toluene to give 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-(N,N-dimethylsulphamoyl)-3-trifluoromethylpyrazole (0.8 g) as light brown crystals, m.p. 177.6-178.6°C.

REFERENCE EXAMPLE 7

5-Amino-4-chlorosulphonyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole used in the above example was prepared as follows:- 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (9.1 g) was added portionwise to stirred cooled chlorosulphonic acid (16.2 ml), keeping the internal temperature below 10°C. The orange solution was stirred at room temperature for 30 minutes, then at 120°C for 5 hours, and poured onto iced water (300 ml) to give a pale brown solid. This solid was filtered, dried, and recrystallised from cyclohexane to give the title compound as yellow crystals.

EXAMPLE 25

Compound No. 46

A solution of 2,6-dichloro-4-trifluoromethyl-phenylhydrazine (3.8 g) and 1,1-dicyano-2-cyclopropyl-2-methoxyethylene (2.23 g) in methanol (30 ml) was stirred and treated with sodium hydride (80%, 30 mg). After 4 hours the solution was evaporated in vacuo and the residue was dissolved in ethyl acetate (40 ml), treated with charcoal and washed with water. The organic phase was evaporated in vacuo, the residual oil was dissolved in petroleum ether and crystals of 5-amino-4-cyano- 3-cyclopropyl-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole, m.p. 197-199°C, were obtained.

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EXAMPLE 26Compounds Nos. 48, 49 and 50

By proceeding in a similar manner to that hereinbefore described in Example 1, but replacing 2,4,6-trichlorophenylhydrazine by 2,6-dichloro-4-trifluoromethylthiophenylhydrazine, there was obtained:-

5 5-Amino-3,4-dicyano-1-(2,6-dichloro-4-trifluoromethylthiophenyl)pyrazole, m.p. 226-227°C, in the form of an off-white solid, after recrystallisation from toluene.

10 By employing 2-chloro-3,5,6-trifluoro-4-trifluoromethylphenylhydrazine there was prepared:-

15 5-Amino-1-(2-chloro-3,5,6-trifluoro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, m.p. 242-243°C, in the form of an orange solid, after recrystallisation from a mixture of ethanol and water.

20 By employing 2,6-dichloro-3,5-difluoro-4-trifluoromethylphenylhydrazine there was prepared:-

25 5-Amino-1-(2,6-dichloro-3,5-difluoro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, m.p. 245-247°C, in the form of an off-white solid.

REFERENCE EXAMPLE 8

2,6-Dichloro-4-trifluoromethylthiophenylhydrazine was prepared by following the procedure of Reference Example 1, by proceeding in a similar manner, but replacing the 2,6-dichloro-4-trifluoromethylaniline by 2,6-dichloro-4-trifluoromethylthianiline.

REFERENCE EXAMPLE 9

2-Chloro-3,5,6-trifluoro-4-trifluoromethyl-phenylhydrazine was prepared as follows:-

5 3-Chloro-2,4,5,6-tetrafluorobenzotrifluoride (12.1 g) and hydrazine hydrate (3.4 g) were heated under reflux with ethanol (50 ml) for 3.5 hours. The mixture was poured onto ice/water mixture (500 ml), stirred, and the product was filtered. After washing with water and drying in a desiccator the title compound was obtained in the form of white crystals, 10 m.p. 91-92°C.

15 By proceeding in a similar manner but replacing 3-chloro-2,4,5,6-tetrafluorobenzotrifluoride by 3,5-dichloro-2,4,6-trifluorobenzotrifluoride there was prepared 2,6-dichloro-3,5-difluoro-4-trifluoromethylphenylhydrazine in the form of pale yellow crystals, m.p. 78-80°C.

EXAMPLE 27

Compound No. 51

20 By proceeding in a similar manner to that hereinbefore described in Example 2, but employing 2,6-dichloro-4-trifluoromethoxyphenylhydrazine there was obtained:-

25 5-Amino-1-(2,6-dichloro-4-trifluoromethoxy-phenyl)-3,4-dicyanopyrazole, m.p. 231-232°C in the form of a brown solid, after recrystallisation from toluene.

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REFERENCE EXAMPLE 10

2,6-Dichloro-4-trifluoromethoxyphenylhydrazine used in the above Example 27 was prepared by following the procedure of Reference Example 1, by proceeding in a similar manner, but replacing the 2,6-dichloro-4-trifluoromethylaniline by 2,6-dichloro-4-trifluoromethoxyaniline. The title compound was obtained as fawn crystals, m.p. 64-65°C.

EXAMPLE 2810 Compounds Nos. 52, 53, 54 and 55

By proceeding in a similar manner to that hereinbefore described in Example 3, but replacing the ethoxyethylenemalononitrile by ethoxypropylene-malononitrile there was prepared:-

15 5-Amino-4-cyano-3-ethyl-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole in the form of white crystals, m.p. 158-160°C, after recrystallisation from a mixture of ethyl acetate and hexane.

20 By proceeding in a similar manner but replacing the ethoxyethylenemalononitrile by ethoxyethylenemethanesulphonylacetone, and by replacing the sodium acetate and glacial acetic acid by ethanol containing 10 mol % of triethylamine at reflux, there was prepared:-

25 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-methylpyrazole in the form of a white solid, m.p. 195°C, after recrystallisation from a mixture of ethyl acetate and hexane.

By proceeding in a similar manner but replacing the ethoxyethylenemalononitrile by ethoxyethylenecyanoacetic acid ethyl ester there was prepared:-

5 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-ethoxycarbonylpyrazole in the form of white crystals, m.p. 115-118°C after recrystallisation from a mixture of toluene and petroleum ether.

10 By proceeding in a similar manner but replacing the ethoxyethylenemalononitrile by ethoxyethylenemethanesulphonylacetone, and by replacing the 2,6-dichloro-4-trifluoromethylphenylhydrazine by 2,6-dichloro-4-trifluoromethoxyphenylhydrazine and by performing the reaction in a 1:1 v/v mixture of ethanol and triethylamine at ambient temperature, there was obtained:-

15 5-Amino-1-(2,6-dichloro-4-trifluoromethoxy-phenyl)-4-methanesulphonyl-3-methylpyrazole, in the form of a fawn solid, m.p. 180-181°C.

REFERENCE EXAMPLE 11

3-Ethoxy-2-methanesulphonylbut-2-ene-nitrile, used in the above Example 28 was prepared as follows:-

20 A mixture of methanesulphonylacetone (200 g), triethylorthoacetate (348 g) and zinc chloride (21 g) was stirred in hexane (1200 ml) with heating under reflux. The distillate was collected via a McIntyre bead, with additional hexane added to the reaction mixture as necessary. Hexane (2800 ml) was collected

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1 during 8 hours. After cooling, the mixture was
2 evaporated in vacuo, and re-evaporated after addition
3 of toluene (100 ml). The residue was dissolved in
4 ethyl acetate and recrystallised from a mixture of
5 ethyl acetate with hexane, twice, to give white
6 crystals, m.p. 99°C, of the title compound.

7 EXAMPLE 29

8 Compounds Nos. 56, 57, 58 and 59

9 By proceeding in a similar manner to that
10 hereinbefore described in Example 4, but replacing the
11 2,6-dichloro-4-trifluoromethylphenylhydrazine by
12 2-chloro-3,5,6-trifluoro-4-trifluoromethylphenyl-
13 hydrazine there was obtained:-

14 5-Amino-1-(2-chloro-3,5,6-trifluoro-
15 4-trifluoromethylphenyl)-4-cyano-3-trifluoromethyl-
16 pyrazole, in the form of white crystals,
17 m.p. 187-189°C, after recrystallisation from toluene.

18 By employing 2,6-dichloro-4-trifluoromethyl-
19 thiophenylhydrazine there was obtained:-

20 5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoro-
21 methylthiophenyl)-3-trifluoromethylpyrazole, in the
22 form of pale yellow crystals, m.p. 133.5-134.5°C, after
23 recrystallisation from hexane.

24 By replacing the 2-chloro-1,1-dicyano-
25 2-trifluoromethylethylene by 2,3-dichloro-1,1-dicyano-
26 3-fluoromethylethylene there was obtained:-

5-Amino-3-chlorofluoromethyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole in the form of a cream solid, m.p. 186-188°C, after recrystallisation from a mixture of toluene and hexane.

By employing 2,6-dichloro-3,5-difluoro-4-trifluoromethylphenylhydrazine there was obtained:-

5-Amino-4-cyano-1-(2,6-dichloro-3,5-difluoro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole in the form of a light brown solid, m.p. 176-177°C.

REFERENCE EXAMPLE 12

Chloro-dicyanoethylene used as starting material in the above Example 29, not hitherto described in the chemical literature was prepared as follows:-

By proceeding in a similar manner to that hereinbefore described in Reference Example 2, but replacing 2-cyano-3-hydroxy-4-chloro-4,4-difluorobut-2-enenitrile sodium salt by 2-cyano-3-hydroxy-4-chloro-4-fluorobut-2-enenitrile sodium salt there was prepared 2-chloro-2-chlorofluoromethyl-1,1-dicyanoethylene as a liquid, b.p. 90°C (46 mmHg).

By proceeding in a similar manner to that hereinbefore described in Reference Example 3, but replacing methyl chlorodifluoroacetate by ethyl chloro-fluoroacetate, there was obtained 2-cyano-3-hydroxy-4-chloro-4-fluorobut-2-enenitrile sodium salt as an orange-red solid.

EXAMPLE 30

Compound No. 60

By proceeding in a similar manner to that hereinbefore described in Example 9, but replacing the triethylorthocformate by triethylorthooacetate there was 5 prepared:-

4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-5-(1-ethoxyethylideneamino)-3-methylpyrazole 10 as a white solid, m.p. 50-53°C, after purification by chromatography on silica (Merck 230-400 mesh, 0.7 kg cm^{-2}) using dichloromethane as eluent.

EXAMPLE 31

Compounds Nos. 61, 62 and 63

By proceeding in a similar manner to that 15 hereinbefore described in Example 10, but replacing the 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-
3,4-dicyanopyrazole by 5-amino-1-(2,6-dichloro-
4-trifluoromethylphenyl)-4-cyano-3-methylpyrazole and acylating with succinyl dichloride there was obtained:-

20 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-
3-methyl-5-succinimidopyrazole in the form of a white
solid, m.p. 202-204°C, after purification by
chromatography on silica (Merck 230-400 mesh, 0.7 kg cm^{-2}) using dichloromethane/ethyl acetate (98:2) as 25 eluent.

By proceeding in a similar manner but replacing the 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole, and employing acetonitrile as solvent for the acylation, there was prepared:-

5-Acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole in the form of a white solid, m.p. 194-195°C, after recrystallisation from toluene.

By proceeding in a similar manner (to Example 10) but replacing the 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-methanesulphonylpyrazole there was prepared 5-acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-methanesulphonylpyrazole in the form of yellow crystals, m.p. 202-203°C.

EXAMPLE 32

Compound No. 64

By proceeding in a similar manner to that hereinbefore described in Example 11, but replacing the 2-chloro-4-trifluoromethylphenylhydrazine by 2,6-dichloro-4-nitrophenylhydrazine there was prepared:-

5-Amino-1-(2,6-dichloro-4-nitrophenyl)-3,4-dicyanopyrazole, in the form of a pale brown solid, m.p. 289-290°C.

EXAMPLE 33

Compounds Nos. 65 and 66

By proceeding in a similar manner to that hereinbefore described in Example 12, but replacing the 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyano-pyrazole and using an appropriate quantity of methyl iodide there was prepared:-

10 1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-methylaminopyrazole in the form of a pale yellow solid, m.p. 165-166°C, after recrystallisation from toluene.

By proceeding as above, but employing ethyl iodide, 15 there was prepared:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-ethylaminopyrazole in the form of an off-white solid, m.p. 245-246°C, after purification by chromatography on silica (Merck 230-400 mesh, 0.7 kg 20 cm^{-2}) using a mixture of ethyl acetate and petroleum ether (15:85).

EXAMPLE 34

Compounds Nos. 67, 68, 69, 70, 71, 72, 73, 74, 75, 76,
77, 78 and 79

By proceeding in a similar manner to that hereinbefore described in Example 14, but replacing the 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole and the trimethyl-acetyl chloride by the following phenylpyrazoles and acylating agents, there were prepared:-

10 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-(N-methyl-N-ethoxycarbonylamino)-3-trifluoromethylpyrazole in the form of a white solid, m.p. 88-90°C, after recrystallisation from hexane, using 4-cyano-1-(2,6-dichloro-4-trifluoro-methylphenyl)-5-methylamino-3-trifluoromethylpyrazole and ethyl chloroformate;

15 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-(N-acetyl-N-trimethylacetylarnino)-3-trifluoro-methylpyrazole in the form of an off-white solid, m.p. 83.5-84°C, after recrystallisation from hexane, using 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-trimethylacetylarnino-3-trifluoromethylpyrazole and acetyl chloride;

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4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-(N-propionyl-N-trimethylacetyl amino)-3-trifluoromethylpyrazole in the form of a white solid, m.p. 56-56.5°C, after recrystallisation from hexane, using 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-trimethylacetyl amino-3-trifluoromethylpyrazole and propionyl chloride;

1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethyl-5-trimethylacetyl amino-pyrazole in the form of a white solid, m.p. 219°C, using 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole and trimethylacetyl chloride;

1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-ethoxycarbonyl amino-4-nitro-3-trifluoromethylpyrazole in the form of pale yellow crystals, m.p. 124°C, using 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole and ethyl chloroformate;

and 3-Chloro-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-5-trimethylacetylaminopyrazole, in the form of a white solid, m.p. 203-204°C;

3-Chloro-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-5-bis(ethoxycarbonyl)aminopyrazole, in the form of an orange crystalline solid, m.p. 67-69°C;

and 3-Chloro-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-5-ethoxycarbonylamino pyrazole, in the form of a yellow solid, m.p. 175-179°C;

5 [The latter three compounds were obtained by reaction of 5-amino-3-chloro-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano pyrazole with the appropriate acyl chlorides]

10 4-Cyano-5-diacetylamino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole in the form of white crystals, m.p. 138-139°C; and 5-(N-Acetyl-N-ethoxycarbonylamino)-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole in the form of a white solid, m.p. 101-102°C;

15 [The above two compounds were obtained by reaction of 5-acetylamino-1-(2,6-dichloro-4-trifluoromethyl)-4-cyano-3-trifluoromethylpyrazole and the appropriate acyl chlorides] and

20 1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-3,4-dicyanopyrazole and 1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-4-methanesulphonyl-3-trifluoromethylpyrazole were prepared in a similar manner to the procedure described in Example 14, but replacing the

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5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyano-pyrazole and by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole respectively. The trimethylacetyl chloride was replaced by the appropriate quantity of ethyl chloroformate (two equivalents) and 2 equivalents of sodium hydride were also used. The products were white crystals with m.p. 74-76°C, and 148-151°C, respectively.

EXAMPLE 35

Compounds Nos. 79 and 80

By proceeding in a similar manner to that hereinbefore described in Example 15, but replacing the 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(cyclopropanecarbonyl)amino-3-trifluoromethylpyrazole by 1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-4-methanesulphonyl-3-trifluoromethylpyrazole there was obtained:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-5-ethoxycarbonylamino-4-methanesulphonyl-3-trifluoromethylpyrazole in the form of a white solid, m.p. 138-141°C.

By proceeding in a similar manner but replacing the 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(cyclopentanecarbonyl)amino-3-trifluoromethylpyrazole by 1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis(ethoxycarbonyl)amino-3,4-dicyanopyrazole there was obtained:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-ethoxycarbonylaminopyrazole in the form of a white solid, m.p. 161-163°C.

10 EXAMPLE 36

Compounds Nos. 81 and 82

By proceeding in a similar manner to that hereinbefore described in Example 18, but replacing N-bromosuccinimide by N-iodosuccinimide there was obtained:-

15 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-iodo-3-trifluoromethylpyrazole, in the form of a white solid, m.p. 129°C.

20 By replacing N-bromosuccinimide by N-iodosuccinimide, and replacing 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole (hereinafter described in Reference Example 13), there was obtained:-

25 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-iodo-3-methylpyrazole in the form of a buff solid, m.p. 108-109°C, after recrystallisation from hexane.

REFERENCE EXAMPLE 13

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methylpyrazole was prepared as follows:-

5-Amino-4-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole (28 g) was heated to 190°C under nitrogen, and maintained at this temperature until gas evolution ceased. After cooling, the title compound was obtained (22 g) as a yellow gum.

10 5-Amino-4-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole used above was prepared by proceeding in a similar manner to Reference Example 6 but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methoxycarbonyl-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethoxycarbonyl-3-methylpyrazole (hereinbefore described in Example 28), and by performing the base hydrolysis at the reflux temperature in ethanol for 13 hours. The title compound was obtained as a white solid, m.p. 183-184°C.

EXAMPLE 37Compounds Nos. 83, 84 and 85

By proceeding in a similar manner to that hereinbefore described in Example 20, but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole, and

replacing the mixture of concentrated sulphuric and fuming nitric acids by concentrated nitric acid alone, there was obtained:-

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-nitropyrazole in the form of orange crystals, m.p. 229-231°C, after recrystallisation from a mixture of toluene and petroleum ether.

By proceeding in a similar manner but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole by 5-acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole, and replacing the mixture of concentrated sulphuric and fuming nitric acids by a mixture of acetic acid and acetic anhydride to which was added fuming nitric acid, there was obtained:-

5-Acetamido-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole in the form of a cream solid, m.p. 194-195°C.

By proceeding in a similar manner but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole by 1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole, and replacing the mixture of concentrated sulphuric and fuming nitric acids by acetic anhydride to which was added fuming nitric acid, there was obtained:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole in the form of an orange solid, m.p. 110-112°C, after recrystallisation from a mixture of toluene and hexane.

REFERENCE EXAMPLE 14

1-(2,6-Dichloro-4-trifluoromethylphenyl)-
3-trifluoromethylpyrazole used in the above Example 37
was prepared by the procedure described in Example 21
5 by replacing 5-amino-4-cyano-1-(2,6-dichloro-
4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by
5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-
3-trifluoromethylpyrazole. The title compound was
obtained as a pale yellow oil.

10 5-Acetamido-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-3-trifluoromethylpyrazole used in the above
Example 37 was prepared by the procedure described in
Example 15, but replacing the 4-cyano-1-(2,6-dichloro-
4-trifluoromethylphenyl)-5-bis(cyclopropanecarbonyl)-
15 amino-3-trifluoromethylpyrazole by 5-bis(acetyl)amino-
1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoro-
methylpyrazole. The title compound was obtained as
white crystals, m.p. 142-144°C, after recrystallisation
from ethyl acetate and hexane.

20 5-Bis(acetyl)amino-1-(2,6-dichloro-
4-trifluoromethylphenyl)-3-trifluoromethylpyrazole,
used above, was prepared by the procedure of Example 19
but replacing 5-amino-4-cyano-1-(2,6-dichloro-
4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by
25 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-
3-trifluoromethylpyrazole, and the ethyl chloroformate
by acetyl chloride. The title compound was obtained
as a white solid, m.p. 130-131°C.

EXAMPLE 38

Compound Nos. 86, 87 and 88

By proceeding in a similar manner to that hereinbefore described in Example 21, but replacing the 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-methanesulphonylpyrazole, there was obtained 1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-methanesulphonylpyrazole in the form of yellow crystals, m.p. 168-169°C.

By proceeding in a similar manner but replacing the 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-fluoropyrazole, there was obtained 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-fluoropyrazole in the form of white crystals, m.p. 120-121°C.

1-(2,6-Dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole was prepared in a similar manner by replacing 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole. The title compound was obtained in the form of white needles, m.p. 154-155°C.

EXAMPLE 39

Compound No. 89

By proceeding in a similar manner to that hereinbefore described in Example 22, but replacing the iodine by anhydrous cupric chloride, and by replacing the chloroform by anhydrous acetonitrile, there was obtained:-

5-Chloro-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-3-trifluoromethylpyrazole in the form of a yellow oil, after purification by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm^{-2}) eluting with a mixture of dichloromethane and hexane (1:2). Infra-Red Absorption bands: 2250, 1495, 1405, 1325, 1160 cm^{-1} (liquid film)

EXAMPLE 40

Compound Nos. 90 and 91

By proceeding in a similar manner to that hereinbefore described in Example 24, but replacing the dimethylamine by the appropriate amines there was prepared the following phenylpyrazoles:-

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-(N-ethylsulphamoyl)-3-trifluoromethyl-pyrazole in the form of a cream solid, m.p. 200°C, after recrystallisation from a mixture of toluene and petroleum ether.

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-(N-methylsulphamoyl)-3-trifluoromethyl-pyrazole in the form of a light brown solid, m.p. 199-200°C, after recrystallisation from toluene.

EXAMPLE 41

Compounds Nos. 92 and 93

Trifluoroacetic anhydride (3.5 ml) was added dropwise to a stirred mixture of 85% w/v hydrogen peroxide solution (0.56 ml) in dichloromethane (15 ml) maintaining at 0-10°C. After warming to 20°C during 5 minutes, a solution of 3-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyanopyrazole (1.0 g; hereinafter described in Reference Example 15) in dichloromethane (10 ml) was added dropwise over 5 minutes. A temperature rise of 10°C was observed during the addition, and the mixture heated under reflux for 1.5 hours. After cooling, the solution was poured into excess water, and the organic solution washed in turn with solutions of sodium bicarbonate and sodium bisulphite. Drying over anhydrous magnesium sulphate, followed by evaporation in vacuo gave a buff solid, which was purified by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm^{-2}) eluting with dichloromethane. The resultant white solid was recrystallised from a mixture of dichloromethane and hexane to give
1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-3-nitropyrazole as white crystals (0.7 g),
m.p. 163-165°C.

By proceeding in a similar manner but replacing 3-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyanopyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole (hereinbefore described in Example 2), there was obtained:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-3,4-dicyano-5-nitropyrazole as orange crystals, m.p. 138-140°C, after recrystallisation from cyclohexane.

REFERENCE EXAMPLE 15

3-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyanopyrazole was prepared as follows:-

A solution of 3-tert-butoxycarbonylaminol-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyanopyrazole (2.8 g) in ethanol (100 ml) was treated with 50% v/v hydrochloric acid (10 ml), and the mixture heated under reflux for 1 hour. After standing overnight at room temperature, sodium carbonate was added to pH 8, and the mixture extracted three times with dichloromethane. The extract was washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a buff solid. Recrystallisation from a mixture of ethyl acetate and petroleum ether gave the title compound (1.4 g) in the form of white crystals, m.p. 159-160°C.

3-tert-Butoxycarbonylaminol-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyanopyrazole was prepared as follows:-

A mixture of 3-carboxy-4-cyano-

1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole (11 g) and phionyl chloride (35 ml) and N,N-dimethylformamide (3 drops) was heated under reflux for 2 hours. The solvent was evaporated in vacuo, and re-evaporated in vacuo after addition of dry toluene (20 ml). The resultant gum was dissolved in dry acetone (50 ml) and stirred, whilst a solution of sodium azide (2.9 g) in water (15 ml) was added during 5 minutes keeping at 10-15°C. After 30 minutes the mixture was poured onto water (250 ml) and extracted with dichloromethane (3 x 80 ml). The combined extract was washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo at equal to or below 40°C to give a buff solid (13 g).

The resulting azide was dissolved in dry toluene (200 ml) and heated under reflux for half-an-hour, with smooth evolution of nitrogen. After cooling, this was treated with tert-butanol (40 g), and the mixture heated under reflux overnight. After evaporation in vacuo, the resulting brown oil (15 g) was purified by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm²) eluting with dichloromethane and ethyl acetate (98:2) to give the title compound (8.0 g) as a white solid, m.p. 154-155°C.

3-Carboxy-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)pyrazole was prepared as follows:-

A suspension of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonylpyrazole

5 (5.0 g) in ethanol (100 ml) was treated with a solution of sodium hydroxide (0.63 g) in water (15 ml) and stirred at room temperature for 1.5 hours. After evaporation in vacuo at equal to or below 40°C, the residue was dissolved in water (150 ml) and extracted 10 with dichloromethane (1 x 100 ml). This extract was back-washed with water (2 x 50 ml), and the combined aqueous solutions brought to pH 1 with dilute hydrochloric acid, and then extracted with ethyl acetate (3 x 50 ml). This extract was dried over 15 anhydrous magnesium sulphate, and evaporated in vacuo to give a buff solid (4.6 g). Recrystallisation from a mixture of toluene and hexane gave the title compound in the form of buff crystals (4.4 g), m.p. 203-205°C.

4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonylpyrazole was prepared by 20 following the method described in Example 21, and replacing 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonylpyrazole. The title compound was 25 obtained in the form of buff crystals, m.p. 198-199°C.

EXAMPLE 42

Compounds Nos. 94, 95 and 96

Silver(I) fluoride (5 g) was added in portions during 40 minutes to a vigorously stirred solution of 1,1-dichloro-2,2-dicyanoethylene in acetonitrile (15 ml), maintained at 0-10°C by external cooling. The stirring was continued at room temperature for 1 hour, and the solid filtered off. The filtrate containing 1,1-difluoro-2,2-dicyanoethylene was stirred and cooled whilst a solution of 2,6-dichloro-4-trifluoromethyl-phenylhydrazine (4.9 g) in acetonitrile (15 ml) was added dropwise at 5°C. After stirring overnight the solid was filtered off and the filtrate evaporated in vacuo to give a dark orange oil (6 g). This was purified by chromatography on silica (Merck, 230-400 mesh, 10 lb in⁻²) eluting with dichloromethane to give a white solid. Recrystallisation from a mixture of cyclohexane and ethyl acetate gave 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-cyano-3-fluoropyrazole (0.9 g) as a white solid, m.p. 193-194°C.

By proceeding in a similar manner but replacing the 2,6-dichloro-4-trifluoromethylphenyl- hydrazine by 2,6-dichloro-4-trifluoromethoxyphenyl- hydrazine and by employing 1,1-dichloro- 2,2-dicyanoethylene instead of 1,1-difluoro- 2,2-dicyanoethylene, and by using diethyl ether as solvent, there was prepared

5-amino-3-chloro-1-(2,6-dichloro-4-trifluoromethoxy-phenyl)-4-cyanopyrazole is the form of a yellow solid, m.p. 175-177°C.

By proceeding as immediately above but replacing the 2,6-dichloro-4-trifluoromethoxyphenylhydrazine by 2,6-dichloro-3,5-difluoro-4-trifluoromethylphenylhydrazine, there was prepared 5-amino-3-chloro-4-cyano-1-(2,6-dichloro-3,5-difluoro-4-trifluoromethylphenyl)pyrazole, in the form of yellow crystals, m.p. 206-208°C.

EXAMPLE 43

Compounds Nos. 97, 98, 99, 100, 101, 102 and 103

A stirred solution of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (1.5 g) in dry tetrahydrofuran cooled to -78°C was treated with a solution of n-butyl lithium (2.6 M in hexane, 1.71 ml) dropwise under nitrogen. The temperature was kept below -65°C during the addition, and the resultant solution kept at -78°C for 1 hour. A solution of trimethylsilyl chloride (0.56 ml) in dry tetrahydrofuran (2 ml) was then added, dropwise, during 2 minutes. The mixture was allowed to reach room temperature over 2 hours, left overnight and evaporated in vacuo to give a pale yellow solid. This was dissolved in dichloromethane, washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo. The product was recrystallised from hexane to

give 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-5-trimethylsilylpyrazole as white crystals, m.p. 108-110°C.

5 By proceeding in a similar manner but replacing the trimethylsilyl chloride by the reagents listed below, the following phenylpyrazoles were obtained:-

10 5-*tert*-Butyldimethylsilyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole in the form of white crystals, m.p. 113-115°C; from *tert*-butyldimethylsilyl chloride.

15 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-methylthio-3-trifluoromethylpyrazole, in the form of a white powder, m.p. 73-74°C; from methylthiocyanate.

20 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethyl-5-trifluoromethylthiopyrazole, in the form of white crystals, m.p. 120-122°C; from bis[trifluoromethyl]disulphide.

25 5-Carboxy-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole, in the form of a white solid, m.p. 177-179°C, by pouring the lithiated pyrazole solution onto a large excess of powdered solid carbon dioxide.

By proceeding in a similar manner but replacing the 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 1-(2,6-dichloro-4-trifluoromethylphenyl)-4-nitro-3-trifluoromethylpyrazole, there was prepared:-

1-(2,6-Dichloro-4-trifluoromethylphenyl)-
4-nitro-3-trifluoromethyl-5-trimethylsilylpyrazole, is
the form of a pale green solid, m.p. 101-103°C.

By proceeding in a similar manner but replacing the

5 4-cyano-1-(2,6-dichloro-4-trifluoro-
methyphenyl)-3-trifluoromethylpyrazole by
4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-
5-methyl-3-trifluoromethylpyrazole (hereinbefore
described in Example 23), there was prepared:-

10 4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-3-trifluoromethyl-5-trimethylsilylmethyl-
pyrazole, in the form of a colourless oil.
Infra-Red Absorption bands: 2250, 1400, 1325, 1260,
1180, 1150, 860cm^{-1} (liquid film)

15 Nuclear Magnetic Resonance: chemical shift (delta) for
 $-\text{Si-CH}_2-$ 2.8ppm in dimethylsulphoxide- D_6 .

EXAMPLE 44

20 Compound No. 104
Sodium methoxide (0.3 g) was added to an ice cold
stirred mixture of 4-cyano-1-(2,6-dichloro-
4-trifluoromethylphenyl)-5-bis(phenoxy carbonyl)amino-
3-trifluoromethylpyrazole (3.1 g) in methanol (30 ml),
and heated under reflux for 2 hours. This was poured
25 onto water (200 ml) and extracted with
dichloromethane. The organic solution was washed with
sodium carbonate solution, then with water, dried over
anhydrous magnesium sulphate, and evaporated in

vacuo. The resultant white solid was 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-methoxy-carbonylamino-3-trifluoromethylpyrazole, m.p. 182-183°C.

REFERENCE EXAMPLE 16

5 4-Cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis-(phenoxycarbonyl)amino-3-trifluoromethylpyrazole used in the above Example 44 was prepared by following the procedure of Example 14, but replacing trimethylacetyl chloride by phenyl chloroformate. The 10 title compound was obtained as a white solid, m.p. 168-169°C.

EXAMPLE 45

Compounds Nos. 105 and 106

15 5-Carbamoyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (3.57 g) was heated to 200°C with phosphorus pentoxide (2.82 g) with stirring. After 3 hours, the cooled product was treated with ice, and extracted with dichloromethane (3 x 50 ml). The organic solution was 20 washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a solid. Recrystallisation from hexane gave 1-(2,6-dichloro-4-trifluoromethylphenyl)-4,5-dicyano-3-trifluoromethylpyrazole in the form of white crystals (1.8 g), 25 m.p. 80°C.

By proceeding in a similar manner but replacing the 5-carbamoyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-3-carbamoyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulph酰pyrazole there was prepared:-

5-Amino-3-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulph酰pyrazole in the form of a white solid, m.p. 214°C.

REFERENCE EXAMPLE 17

10 5-Carbamoyl-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole used in the above Example 45, was prepared as follows:-

15 5-Carboxy-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (6.0 g; hereinbefore described in Example 43) was added to thionyl chloride (30 ml) and the stirred solution heated to reflux for 4 hours. The solvent was evaporated in vacuo, and re-evaporated after addition of dry toluene (30 ml). The resultant orange oil was

20 dissolved in dry ether (10 ml) and added dropwise to a stirred solution of ammonia (0.88, 20 ml) cooled by an ice bath. After stirring overnight, water (150 ml) was added, and the mixture extracted with dichloromethane (3 x 50 ml). The combined extract was washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a white solid (7.0 g). Recrystallisation from a mixture of ethyl acetate and petroleum ether gave the title

25

compound (4.0 g), in the form of white crystals, m.p. 180-181°C.

5-Amino-3-carbamoyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonylpyrazole

used in the above Example 45 was prepared by the same procedure, but by replacing the 5-carboxy-4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole by 5-amino-3-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonylpyrazole.

10 The title compound was obtained in the form of an off-white solid, m.p. 223-224°C.

5-Amino-3-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonylpyrazole used above was prepared as follows:-

15 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonyl-4-methanesulphonylpyrazole (8.15 g) was added to stirred 80% sulphuric acid (80 ml), and heated at 100°C for 5 hours. After cooling, the solution was poured onto ice, the solid filtered off and dried over phosphorus pentoxide in a vacuum desiccator. Recrystallisation from a mixture of methanol and petroleum ether gave the title compound as a white solid, m.p. 203-205°C.

20 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-ethoxycarbonyl-4-methanesulphonylpyrazole, used above, was prepared by the procedure of Reference Example 5, by replacing malononitrile by

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methanesulphonylacetonicitrile. The title compound was obtained in the form of a white solid, m.p. 255°C, after recrystallisation from ethanol.

EXAMPLE 46

5 Compound No. 107

A solution of methylmagnesium iodide [prepared from magnesium (0.25 g) and methyl iodide (1.5 g) in diethyl ether (25 ml)], was treated with a solution of 1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-cyano-3-trifluoromethylpyrazole (2 g) in diethyl ether (20 ml), dropwise. The resulting pale yellow solution was refluxed for 24 hours, cooled, and treated with hydrochloric acid (2N, 10 ml). After stirring for 0.5 hour at room temperature, the reaction mixture was diluted with ether (50 ml). The ethereal extract was washed with water (50 ml), dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a yellow gum. This was purified by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm^{-2}) eluting with a mixture of dichloromethane and petroleum ether (4:1) to give 4-acetyl-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole as a white solid, m.p. 134°C.

EXAMPLE 47

25 Compounds Nos. 108 - 116

A stirred solution of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methylthio-3-trifluoromethyl-pyrazole (1.0 g) in chloroform (40 ml) was treated

with *m*-chloroperbenzoic acid (0.42 g), portionwise at room temperature. After stirring for 6 hours, the solution was diluted with dichloromethane and washed in turn with sodium sulphite solution, sodium

5 hydroxide solution, and water. The solution was dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a yellow oil.

Purification by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) eluting with

10 dichloromethane-ethylacetate (4:1) gave 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methylsulphinyl-3-trifluoromethylpyrazole in the form of a white solid, m.p. 142-145°C with decomposition.

By proceeding in a similar manner but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methylthio-3-trifluoromethylpyrazole by the appropriate alkylthio phenylpyrazoles there were prepared:-

20 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylsulphinyl-3-trifluoromethylpyrazole in the form of a white solid, m.p. 170°C from 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylthio-3-trifluoromethylpyrazole.

25 5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylsulphinyl-3-methylpyrazole in the form of a buff solid, m.p. 157-158°C from 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylthio-3-methylpyrazole.

5-Amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylsulphinylphenyl)-3-trifluoromethylpyrazole, in the form of an orange solid, m.p. 76°C, from 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylthiophenyl)-3-trifluoromethylpyrazole.

4-Cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-methylsulphinyl-3-trifluoromethylpyrazole, in the form of white crystals, m.p. 97-98°C, from 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-methylthio-3-trifluoromethylpyrazole.

By proceeding in a similar manner but
replacing 5-amino-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-4-methylthio-3-trifluoromethylpyrazole by the
appropriate alkylthiophenylpyrazoles, and employing 2
molar equivalents of α -chloroperbenzoic acid there was
prepared:-

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-
phenyl)-4-ethylsulphonyl-3-trifluoromethylpyrazole, in
the form of white crystals, m.p. 206-207°C, from
5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-
4-ethylthio-3-trifluoromethylpyrazole.

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-ethylsulphonyl-3-methylpyrazole, in the form of a white solid, m.p. 193°C, from 5-amino-
1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylthio-3-methylpyrazole.

By proceeding in a similar manner but replacing the 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methylthio-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-a-propylthio-3-methylpyrazole, there was obtained 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-propanesulphonylpyrazole in the form of a white solid, m.p. 145.5-147°C.

By proceeding in a similar manner there was prepared:-

5-Amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-trichloromethanesulphonyl-3-methylpyrazole in the form of a pale pink solid, m.p. 183-184°C, from 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-trichloromethylthio-3-methylpyrazole.

EXAMPLE 48

Compounds Nos. 117, 118 and 119

A mixture of bis[5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole-4-yl]-disulphide (4.0 g), sodium dithionite (2.02 g) and sodium hydroxide (0.46 g) was stirred and heated under reflux in a mixture of ethanol and water (60 ml, 1:1) for 4 hours. The cooled yellow solution was treated with ethyl iodide (2.17 g) and the mixture stirred and heated under reflux for 2 hours. After evaporation in vacuo, the yellow gum was dissolved in ether (100 ml), washed with water, dried over anhydrous magnesium sulphate, and re-evaporated in vacuo. The

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residual gum was purified by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm^{-2}) eluting with dichloromethane, to furnish 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylthio-3-methylpyrazole in the form of a white solid, m.p. 117°C, after recrystallisation from hexane.

By proceeding in a similar manner, but replacing the ethyl iodide by methyl iodide there was prepared:-

5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-methylthiopyrazole, in the form of a white solid, m.p. 112°C, after recrystallisation from hexane.

By proceeding in a similar manner but replacing the sodium hydroxide by sodium carbonate, and the methyl iodide by *n*-propyl iodide, there was obtained 5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-*n*-propylthio-3-methylpyrazole in the form of a white solid, m.p. 100-102°C.

REFERENCE EXAMPLE 18

Bis[5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methylpyrazole-4-yl]disulphide was prepared as follows:-

A solution of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-thiocyanoypyrazole (3.0 g; herein after described in Example 50) in a mixture of ethanol and water (1:1, 100 ml) was acidified by the addition of hydrochloric acid (10 N).

20 ml). The mixture was heated under reflux for 8 hours, concentrated to half volume in vacuo, cooled in an ice bath, and sodium hydroxide solution added until the pH reached 9-10. The precipitated product was filtered, washed with water, and dried in vacuo to furnish the title compound (2.66 g) as an amorphous yellow powder, m.p. 211-213°C.

EXAMPLE 49

Compounds Nos. 120 and 121

10 A solution of ethyl magnesium bromide, prepared from magnesium (0.57 g) and ethyl bromide (2.6 g) in dry diethyl ether (25 ml), was added dropwise to a stirred solution of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-thiocyanato-3-trifluoromethylpyrazole (5.0 g) in dry ether (50 ml) at -20°C. After stirring for a further 2 hours at room temperature, water (130 ml) was carefully added, and stirring maintained for 0.25 hour. The ether layer was separated, dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a yellow gum. Purification by chromatography on silica (Merck, 230-400 mesh, 0.7 kg cm⁻²) eluting with dichloromethane-petroleum ether (1:1) gave a product, which recrystallised from hexane to furnish 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-ethylthio-3-trifluoromethylpyrazole, in the form of white needles, m.p. 116-116.5°C.

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By proceeding in a similar manner, but replacing the ethyl magnesium iodide by methyl magnesium iodide there was obtained:-

5 5-Amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methylthio-3-trifluoromethylpyrazole, in the form of a white solid, m.p. 108°C, after rectyscallisation from hexane.

EXAMPLE 50

Compounds Nos. 122 and 123

10 A stirred mixture of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (0.7 g) and potassium thiocyanate (0.55 g) in methanol (15 ml) was treated with a solution of bromine (0.3 g) in methanol (2 ml) at 0-5°C. Stirring was maintained 15 at this temperature for 1.5 hours, and the mixture was poured onto ice water, and brought to pH 9 by the addition of sodium carbonate. The product was filtered, washed with water and dried. Purification by chromatography on silica (Merck, 230-400 mesh, 0.7 20 kg cm²) eluting with dichloromethane gave 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-thiocyanato-3-trifluoromethylpyrazole, in the form of a white solid, m.p. 49-50°C.

By proceeding in a similar manner but replacing 5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-trifluoromethylpyrazole by 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methylpyrazole there was obtained 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-thiocyanatopyrazole in the form of a white solid, m.p. 133.5°C, after recrystallisation from a mixture of hexane and ethyl acetate.

10

EXAMPLE 51

Compound No. 124

By proceeding in a similar manner to that hereinbefore described in Example 4, but replacing the 2,6-dichloro-4-trifluoromethylphenylhydrazine by 2,6-dichloro-4-methanesulphonylphenylhydrazine there was obtained:-

5-Amino-4-cyano-1-(2,6-dichloro-4-methanesulphonylphenyl)-3-trifluoromethylpyrazole in the form of white crystals, m.p. 270-272°C.

20

REFERENCE EXAMPLE 19

By proceeding in a similar manner to that hereinbefore described in Reference Example 1, but replacing the 2,6-dichloro-4-trifluoromethylaniline by 2,6-dichloro-4-methanesulphonylaniline, there was prepared:-

2,6-Dichloro-4-methanesulphonylphenylhydrazine in the form of white crystals, m.p. 163-166°C.

EXAMPLE 52

Compound No. 125

To a stirred ice cold solution of 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-pyrazole (2.0 g) in chloroform (40 ml) and pyridine (0.51 g) was added dropwise a solution of trichloromethanesulphenyl chloride (1.2 g) in chloroform (10 ml). The resulting brown solution was stirred at 0°C for 2 hours, then at room temperature for 2 hours. A further addition of trichloromethanesulphenyl chloride (0.5 g) was made and the mixture stirred for 2 hours at room temperature. Water (100 ml) and dichloromethane (100 ml) was then added and the organic layer washed with water (1 x 100 ml), dried over anhydrous magnesium sulphate and evaporated in vacuo to give a yellow gum (2.9 g). This was purified by chromatography on silica (Merck, 100-230 mesh, 0.7 kg cm⁻²) eluting with dichloromethane-petroleum ether (3:2) to give a white solid (0.98 g). Recrystallisation from hexane gave 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-methyl-4-trichloromethylthiopyrazole in the form of white crystals, m.p. 156°C.

EXAMPLE 53

Compound No. 126

m-Chloroperbenzoic acid (2.1 g) was added to a solution of 5-amino-4-cyano-1-(2,6-dichloro-4-trifluoromethylthiophenyl)-3-trifluoromethylpyrazole (2.3 g) in dichloromethane (20 ml) cooled to 10°C. After stirring overnight at room temperature, the solution was heated under reflux for 4 hours, cooled, and a further addition of m-chloroperbenzoic acid (2.1 g) made. The mixture was stirred at room temperature for 4 hours and heated under reflux for 4 hours. The cooled solution was washed with sodium bicarbonate solution (20 x 20 ml), then with water (2 x 20 ml), dried over anhydrous magnesium sulphate, filtered, and evaporated in vacuo to give an orange solid. Purification by chromatography on silica (Merck, 100-230 mesh, 0.7 kg cm⁻²) eluting with ethyl acetate-petroleum ether (1:9) gave 4-cyano-1-(2,6-dichloro-4-trifluoromethanesulphonylphenyl)-5-nitro-3-trifluoromethylpyrazole (0.5 g) in the form of an orange solid, m.p. 168-169°C.

EXAMPLE 54

Compound No. 127

To a stirred solution of diethylaminosulphur trifluoride (1.5 g) in dichloromethane (13 ml) cooled to -70°C, was added dropwise under nitrogen a solution of 1-(2,6-dichloro-4-trifluoromethylphenyl)-4-formyl-3-trifluoromethylpyrazole (3.1 g) in dichloromethane (17 ml). After 1 hour at -70°C, the mixture was allowed to stand at room temperature overnight, then poured onto excess iced water. Extraction with dichloromethane gave a solution which was washed with water (2 x), dried over anhydrous magnesium sulphate and evaporated in vacuo to give a brown oil (3.26 g). Purification by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm^{-2}) eluting with hexane-ethyl acetate (5:1) gave 1-(2,6-dichloro-4-trifluoromethylphenyl)-4-difluoromethyl-3-trifluoromethylpyrazole (1.15 g) [from ethyl acetate-hexane] in the form of a pale yellow solid, m.p. 88-90°C.

20

REFERENCE EXAMPLE 20

A mixture of 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (5.0 g; hereinbefore described in Example 21) and formic acid (120 ml) was treated with Raney nickel (5.1 g) and the mixture heated under reflux overnight. After cooling, the mixture was filtered, and the filtrate diluted with water (900 ml) and extracted with dichloromethane (4 x 100 ml). The

combined extract was washed with sodium bicarbonate solution (2 x), then with water (1 x), dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a brown solid (3.7 g), m.p. 80-82°C. This was 1-(2,6-dichloro-4-trifluoromethylphenyl)-4-formyl-3-trifluoromethylpyrazole.

EXAMPLE 55

Compound No. 128

To a stirred solution of 5-amino-4-carboxy-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-trifluoromethylpyrazole (15.0 g; hereinbefore described in Reference Example 6) in dry tetrahydrofuran (50 ml) was added, under nitrogen, a solution of borane-tetrahydrofuran complex (1 Molar, 27.5 g) during 10 minutes keeping at -20°C. The solution was allowed to reach room temperature and stirred overnight. A further addition of the borane was made (10 ml), and the solution heated under reflux overnight. After cooling, a further addition of the borane (20 ml) was made, and the solution again heated under reflux for 4 hours. After cooling, sodium hydroxide (6 N) solution was added to pH 11, and the solution extracted with dichloromethane. The organic phase was washed with water, dried over anhydrous magnesium sulphate, and evaporated in vacuo to give a brown oil. Purification by chromatography on silica (Merck, 40-230 mesh, 0.7 kg cm⁻²) eluting with

hexane-ethyl acetate (2:1) gave 5-amino-
1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methyl-
3-trifluoromethylpyrazole (2.0 g), from
toluene-hexane, m.p. 97-100°C, in the form of white
5 crystals.

10

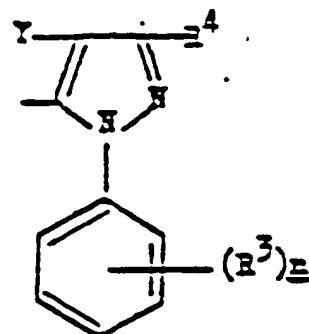
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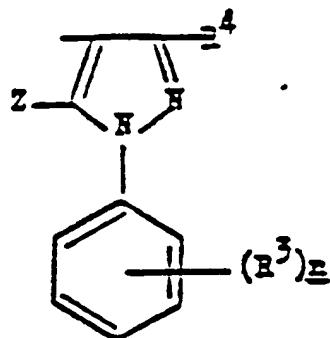
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In the following formulae it is to be understood that:

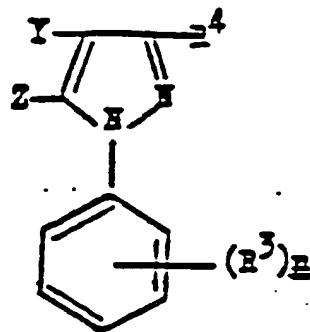
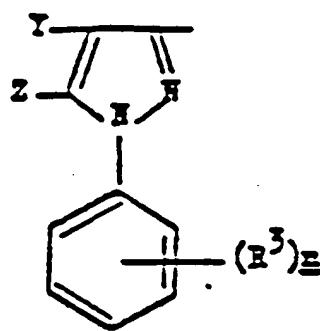
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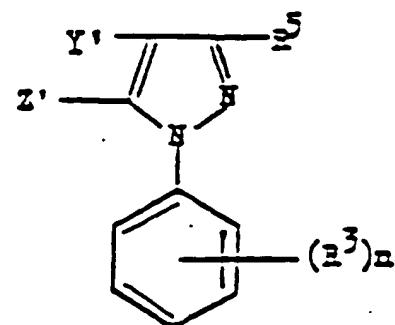


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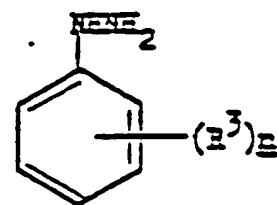


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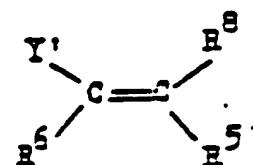




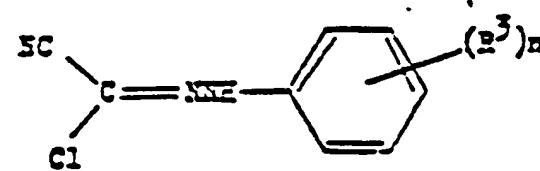
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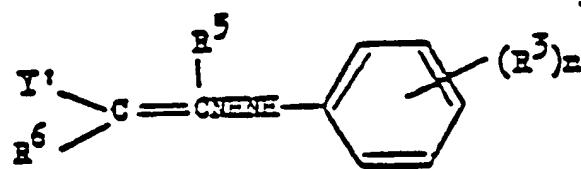
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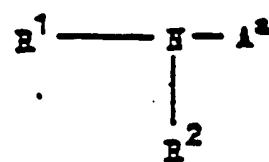
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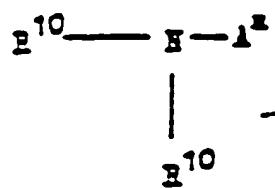
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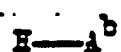
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VI



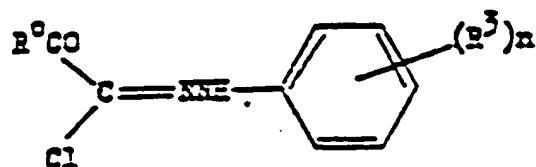
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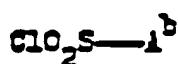
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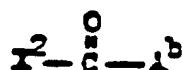
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XIV



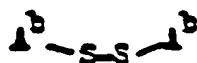
XV



XVI



X C



XVII



XVIII



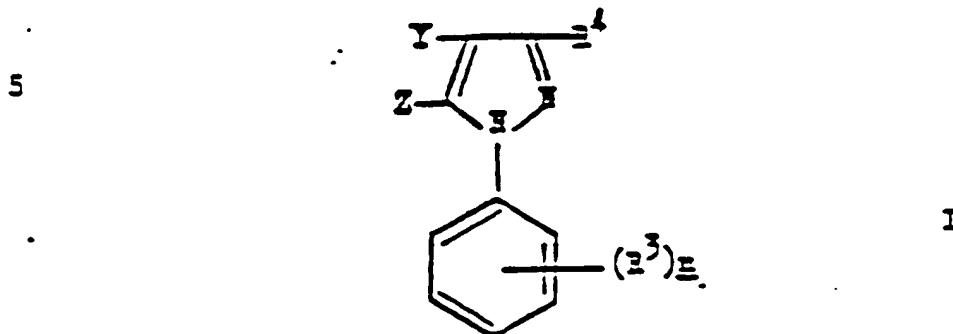
XIX



X

CLAIMS

1. A method for the control of arthropod, plant nematode or helminth pests at a locus which comprises treatment of the locus with an effective amount of a compound of the general formula:



wherein Y represents a halogen atom, the cyano or nitro group or a group RSO_2 , RSO or RS in which R represents a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which is unsubstituted or substituted 10 by one or more halogen atoms, a cycloalkyl group containing from 3 to 5 carbon atoms, a straight- or branched-chain alkenyl group containing from 2 to 6 carbon atoms, the thiocyanato group, the sulphamoyl group which is unsubstituted or substituted by one or two 15 straight- or branched-chain alkyl groups which may be the same or different and contain from 1 to 6 atoms, the carbamoyl group which may be unsubstituted or substituted by one or two straight- or branched-chain alkyl groups which may be the same or different and contain from 1 20 to 6 carbon atoms, a straight- or branched-chain

alkoxycarbonyl group containing from 2 to 7 carbon atoms, a straight- or branched-chain alkanoyl group containing from 2 to 7 carbon atoms, or a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, Z represents the hydrogen atom, or the amino group $-NR^1R^2$ wherein R^1 and R^2 , which may be the same or different, each represents a hydrogen atom or a straight- or branched-chain alkyl group (containing 10 from 1 to 6 carbon atoms, and which is unsubstituted or substituted by straight- or branched-chain alkoxycarbonyl of 2 to 5 carbon atoms), cycloalkyl group containing from 3 to 6 carbon atoms, formyl group, straight- or branched-chain alkanoyl group (which contains from 2 to 7 carbon atoms or together form a 5 or 6 membered cyclic imide with the nitrogen atom to which they are attached and themselves are unsubstituted or substituted by one or more halogen atoms) or cycloalkylcarbonyl group (which contains from 4 to 7 carbon atoms) or straight- or branched-chain alkoxy carbonyl group (which contains from 2 to 7 carbon atoms and themselves are unsubstituted or substituted by one or more halogen atoms), or Z represents a straight- or branched-chain alkylsulphenylamino group containing from 1 to 4 carbon atoms, a straight- or branched-chain alkoxyethyleneamino group containing from 2 to 5 carbon atoms which is unsubstituted or substituted on methylene by a straight-

or branched-chain alkyl group containing from 1 to 4 carbon atoms, or represents a halogen atom, a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms, the carboxy group, or a straight- or 5 branched-chain alkylthio, alkylsulphinyl or alkylsulphonyl group containing from 1 to 6 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, or represents a straight- or branched-chain trialkylsilylmethyl group containing from 10 1 to 6 carbon atoms in each alkyl group which may be the same or different, a trialkylsilyl group containing from 1 to 6 carbon atoms in each alkyl group which may be the same or different or the cyano or nitro group, R³ represents a halogen atom, a straight- or branched-chain 15 alkyl or alkoxy group containing from 1 to 4 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, a straight- or branched-chain alkylthio or alkylsulphinyl group containing from 1 to 4 carbon atoms which is substituted by one or more halogen atoms, the 20 nitro or cyano group or a straight- or branched-chain alkylsulphonyl group containing from 1 to 4 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, and R⁴ represents a halogen atom, a cyano or nitro group or a straight- or branched-chain alkyl 25 group containing from 1 to 4 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, or a cycloalkyl group containing from 3 to 6

carbon atoms, and n represents an integer from 1 to 5 inclusive, and, when Z represents a carboxy group, salts thereof with pesticidally-acceptable bases, provided that R^4 , Y and Z do not simultaneously represent three groups of the same genus selected from the genera (i) nitro, (ii) cyano, (iii) halogen and (iv) unsubstituted alkyl.

2. A method according to claim 1 wherein in general formula I (R^3) $_n$ represents 2,4,6-trichloro-, 2,3,5,6-tetrachloro, 2-chloro-4-trifluoromethyl, 10 2,3,5,6-tetrafluoro-4-trifluoromethyl, 2,6-dichloro-4-trifluoromethylthio, 2-chloro-3,5,6-trifluoro-4-trifluoromethyl, 2,6-dichloro-3,5-difluoro-4-trifluoromethyl, 2,6-dichloro-4-nitro, 15 2,6-dichloro-4-trifluoromethylsulphanyl, 2,6-dichloro-4-methanesulphonyl or 2,6-dichloro-4-trifluoromethanesulphonyl substitution.

3. A method according to claim 1 wherein in general formula I (R^3) $_n$ represents 2,6-dichloro-4-trifluoromethyl or 2,6-dichloro-4-trifluoromethoxy substitution.

4. A method according to any one of claims 1 to 3 wherein the compound of general formula I is 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3,4-dicyanopyrazole, 25 5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole, 4-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-5-bis

(ethoxycarbonyl) amino-3-trifluoromethylpyrazole,

5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-

methanesulphonyl-3-methylpyrazole,

1-(2,6-dichloro-4-trifluoromethylphenyl)-5-

5 ethoxycarbonylamino-4-nitro-3-trifluoromethylpyrazole,

5-amino-3-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-

4-methanesulphonylpyrazole, or

5-amino-1-(2,6-dichloro-4-trifluoromethylphenyl)-3-

methyl-4-methylthiopyrazole.

10 5. An arthropodicidal, plant nematocidal or anthelmintic composition which comprises at least one compound of general formula I, or a pesticidally acceptable salt thereof, in association with one or more compatible diluents or carriers with the provisos that

15 (1) when the composition comprises a single compound of

general formula (I) wherein R⁴ and Z both represent methyl, Y represents thiocyanato and (R³)_n represents 2-, 3- or 4-nitro, 4-methyl, 4-chloro or 2,4-dinitro substitution; or R⁴ represents methyl, Y represents 5 cyano, Z represents unsubstituted amino and (R³)_n represents 4-chloro, 2,4-dichloro, 3,4-dichloro, 3-chloro-4-methyl or 2-methyl-4-chloro substitution, the composition is not an association of a single compound of general formula I alone with water or a common organic solvent; (2) when the composition comprises a single compound of general formula (I) wherein R⁴ represents methyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n represents 3- or 4-fluoro substitution; or R⁴ represents ethyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n represents 3- or 4-chloro, 2-, 3- or 4-fluoro or methyl, 3-bromo or 3-nitro substitution; or R⁴ represents propyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n represents 3-fluoro substitution; or R⁴ represents methyl, Y represents sulphamoyl, Z represents chloro and (R³)_n represents 4-chloro substitution; the composition comprises an agriculturally acceptable surface active

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agent or a feedstuff; (3) when R⁴ represents methyl, Y represents nitro, and Z represents chloro or R⁴ represents chloro, Y represents nitro, and Z represents methyl and (R³)_n represents 4-nitro, the composition 5 comprises a pharmaceutically acceptable adjuvant or a feedstuff or is substantially sterile and pyrogen-free or is in unit dosage form; and (4) excluding compositions comprising

1-(4-nitrophenyl)-3-nitro-4-pyrazole-carbonitrile or 10 carboxamide.

6. A compound of general formula I, wherein the various symbols are as defined in claim 1, and salts thereof, with the exclusion of the compounds wherein: R⁴ and Z both represent methyl, Y represents thiocyanato and 15 (R³)_n represents 2-, 3- or 4-nitro, 4-methyl, 4-chloro or 2,4-dinitro substitution; R⁴ represents methyl, Y represents cyano, Z represents unsubstituted amino and (R³)_n represents 4-chloro, 2,4-dichloro, 3,4-dichloro, 3-chloro-4-methyl or 2-methyl-4-chloro substitution; R⁴ 20 represents methyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and (R³)_n represents 3- or 4-fluoro substitution; R⁴ represents ethyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and

$(R^3)_n$ represents 3- or 4-chloro, 2-, 3- or 4-fluoro or methyl, 3-bromo or 3-nitro substitution;
R⁴ represents propyl, Y represents cyano or CONH₂, Z represents unsubstituted amino and $(R^3)_n$ represents 5 3-fluoro substitution; R⁴ represents methyl, Y represents sulphamoyl, Z represents chloro and $(R^3)_n$ represents 4-chloro substitution; R⁴ represents methyl, Y represents nitro, and Z represents chloro or R⁴ represents chloro, Y represents nitro, and Z represents methyl and $(R^3)_n$ 10 represents 4-nitro; and R⁴ represents nitro, Y represents cyano or CONH₂, Z represents hydrogen and $(R^3)_n$ represents 4-nitro substitution.

7. A compound according to 6 wherein $(R^3)_n$ represents 2,4,6-trichloro, 2,3,5,6-tetrachloro,

15 2-chloro-4-trifluoromethyl,
2,3,5,6-tetrafluoro-4-trifluoromethyl,
2,6-dichloro-4-trifluoromethylthio,
2-chloro-3,5,6-trifluoro-4-trifluoromethyl,
2,6-dichloro-3,5-difluoro-4-trifluoromethyl,
20 2,6-dichloro-4-nitro,
2,6-dichloro-4-trifluoromethylsulphanyl,
2,6-dichloro-4-methanesulphonyl or

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2,6-dichloro-4-trifluoromethanesulphonyl substitution.

8. A compound according to claim 6 wherein $(R^3)_n$ represents 2,6-dichloro-4-trifluoromethyl or 2,6-dichloro-4-trifluoromethoxy substitution.

5 9. A compound according to claim 6 which is

5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3,4-dicyanopyrazole,

5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methanesulphonyl-3-trifluoromethylpyrazole,

10 4-cyano-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-5-bis(ethoxycarbonyl)amino-3-trifluoromethylpyrazole,

5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-4-methanesulphonyl-3-methylpyrazole,

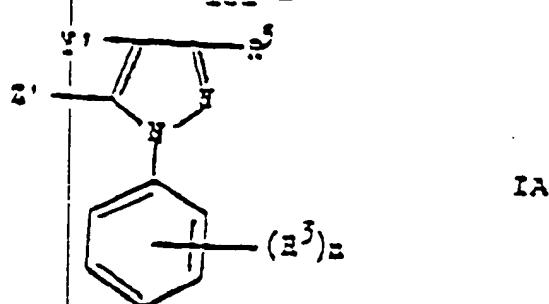
1-(2,6-dichloro-4-trifluoromethylphenyl)-5-

15 ethoxycarbonylamino-4-nitro-3-trifluoromethyl-pyrazole,
5-amino-3-cyano-1-(2,6-dichloro-4-trifluoromethylphenyl)-4-methane sulphonylpyrazole or
5-amino-1-(2,6-dichloro-4-trifluoromethyl-phenyl)-3-methyl-4-methylthiopyrazole.

20 10. A process for the preparation of a compound of general formula I depicted in claim 1 wherein the various symbols are as defined in claim 6 which comprises:

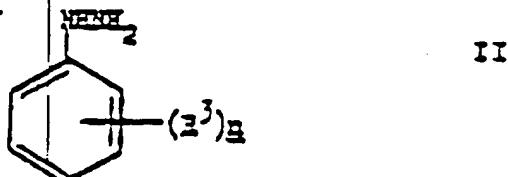
(A) when the compound of general formula I conforms to the general formula:

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wherein Y' represents the cyano or nitro group or a group RSO₂, RSO or RS, wherein R is as defined in claim 1, a straight- or branched-chain alkoxy carbonyl group 5 containing from 2 to 7 carbon atoms, or a straight- or branched-chain alkyl group containing from 1 to 6 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, Z' represents the unsubstituted amino group or a straight- or branched-chain alkyl group 10 containing from 1 to 4 carbon atoms, and R⁵ represents a fluorine, chlorine or bromine atom, the cyano group or a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms which is unsubstituted or substituted by one or more halogen atoms, or a cycloalkyl group 15 containing from 3 to 6 carbon atoms,

(1) the reaction of a compound of the general formula:

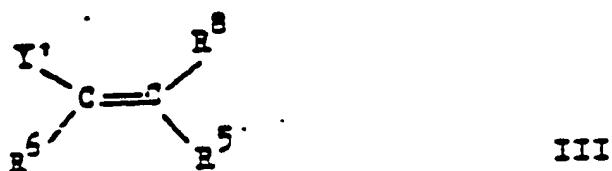


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wherein R^3 and n are as defined in claim 1, or an acid addition salt thereof with (1), when R^5 in the compound of general formula IA represents a fluorine, chlorine or bromine atom, an optionally halogenated straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms or a cycloalkyl group containing from 3 to 6 carbon atoms, a compound of the general formula:



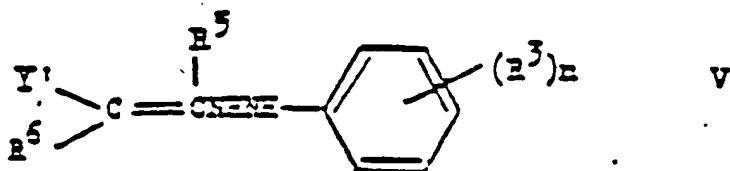
wherein Y' and R^5 are as hereinbefore defined, R^6 represents the cyano group or a straight- or branched-chain alkanoyl group containing from 2 to 5 carbon atoms and R^8 represents a straight- or branched-chain alkoxy group containing from 1 to 4 carbon atoms, the hydroxy group or a fluorine, chlorine or bromine atom, or

(2) when R^5 in the compound of general formula IA represents the cyano group, Y' represents the cyano group and Z' represents the unsubstituted amino group, with tetracyanoethylene; optionally isolating, before its cyclisation to the compound of general formula IA, the intermediate of general formula:

acid

- 183 -

mpour
fine
or
carb
; car

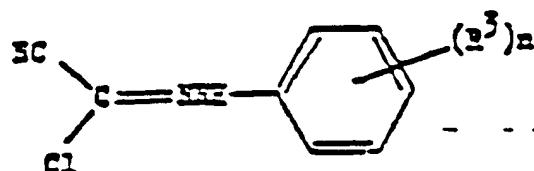


wherein R^3 , n , R^5 , R^6 and Y' are as hereinbefore defined, formed by reaction of a compound of general formula II with a compound of general formula III or 5 tetracyano-ethylene;

(B) when Z' in general formula IA represents the unsubstituted amino group, by reacting a compound of general formula $\text{Y}'\text{CH}_2\text{CN}$ with a compound of general formula II in the presence of a compound of the general formula $\text{R}^7\text{C}(\text{R}^0)_3$ wherein R^7 represents a straight- or branched-chain alkyl group containing from 1 to 4 carbon atoms which is unsubstituted or substituted by one or more halogen atoms or a cycloalkyl group containing from 3 to 6 carbon atoms and R^0 represents an alkoxy group in 15 an inert organic solvent at a temperature from ambient to reflux;

(C) when Z' in general formula IA represents the

unsubstituted amino group and R^5 represents the cyano group by the reaction of a compound of the general formula



IV

5 wherein R^3 and n are as defined in claim 1, with a compound of general formula $Y'CH_2CN$, wherein Y' is as hereinbefore defined; and preparing other compounds of general formula I by conversion, as hereinbefore described, of one or more 10 substituents Y , Z , R^3 and R^4 , or substituents corresponding thereto, into substituents Y , Z , R^3 and I as defined in claim 1; and, when Z represents the carboc group, optionally converting a compound of general formula I into a salt thereof.

15 11. Intermediate compounds of general formula I wherein Y represents the hydrogen atom, the formyl or carboxy group, a straight- or branched-chain alkanoyl group containing from 2 to 6 carbon atoms which is substituted by one or more halogen atoms, the dithio 20 group (which joins two pyrazole rings), the amino group

the $-SO_2Cl$ group, a straight- or branched-chain carboxyalkyl group containing from 2 to 6 carbon atoms, z represents the carbamoyl group or a straight- or branched-chain alkoxy carbonyl group containing from 2 to 5-7 carbon atoms or the diphenoxycarbonylamino group, $(R^3)_n$ substitution is as defined in claim 2 or 3 or R^4 represents the amino, hydroxymethyl, carboxy or carbamoyl group or a straight- or branched-chain alkoxy carbonyl or alkoxy carbonylamino group containing from 2 to 7 carbon atoms.

12. A compound of general formula I as defined in claim 1, or a pesticidally acceptable salt thereof, for use in the manufacture of a medicament for the treatment of an arthropod or helminth infection.

INTERNATIONAL SEARCH REPORT

International Application No. PCT/GB 86/00781

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all)

According to International Patent Classification (IPC) or to both National Classification and IPC

IPC: 4 A 01 N 43/56; A 61 K 31/415; C 07 D 231/38; A 01 N 55/00
C 07 F 7/08; C 07 D 231/44; C 07 D 231/40; C 07 D 231/18

II. FIELDS SEARCHED

Minimum Documentation Searched *

Classification System	Classification Symbols
IPC ⁴	A 01 N; A 61 K; C 07 D; C 07 F

Documentation Searched other than Minimum Documentation
to the Extent that such Documents are Included in the Fields Searched *

III. DOCUMENTS CONSIDERED TO BE RELEVANT*

Category *	Citation of Document, ** with indication, where appropriate, of the relevant passages in	Relevant to Claim No. 13
P,X	EP, A2, 0201852 (BAYER) 20 November 1986 see page 2, line 30 - page 4, line 19; page 9, line 9 - page 12, line 2; page 23, line 3 - page 33, line 24; page 38, line 26 - page 40, line 28; page 50, line 1 - page 54, line 17; page 75, compound 66; page 90, compounds 192,193; claims --	1-12
P,X	EP, A1, 0200872 (BAYER) 12 November 1986 see page 55, line 21 - page 64, line 19; page 68, line 28 - page 72, line 27; claims --	1-3,5-8, 10-12
A	Journal of Heterocyclic Chemistry, Volume 12, no. 12, December 1975, Provo, Utah, US), P.L. Southwick et al.: "Preparation of 4,6-Diaminopyrazolo(3,4-d)pyri- midines with variations in substitution at the 1- and 3-positions (1)", pages 1199-1205 see tables IV,V cited in the application	1-12

* Special categories of cited documents: *

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** document referring to an oral disclosure, use, exhibition or other means

** document published prior to the International filing date but later than the priority date claimed

** later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

** document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step

** document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

** document member of the same patent family

IV. CERTIFICATION

Date of the Actual Completion of the International Search

Date of Mailing of this International Search Report

21 MAY 1987

International Searching Authority

Signature of Authorized Officer

EUROPEAN PATENT OFFICE

M. VAN MOL

INTERNATIONAL SEARCH REPORT

International Application No. PCT/GB 86,

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all)

According to International Patent Classification (IPC) or to both National Classification and IPC

IPC: C 07 D 231/16; C 07 D 231/14; C 07 D 231/12;
C 07 D 403/04

II. FIELDS SEARCHED

Minimum Documentation Searched*

Classification System:

Classification Symbols

IPC⁴Documentation Searched other than Minimum Documentation
to the Extent that such Documents are Included in the Fields Searched*

III. DOCUMENTS CONSIDERED TO BE RELEVANT*

Category* | Citation of Document, ** with indication, where appropriate, of the relevant passages | Relevant to Claim No. 18

A : US, A, 3760084 (J.W. MARSICO, Jr. et al.)
18 September 1973
see claims
cited in the application | 1-12

A : US, A, 4145554 (R.G. JONES) 20 March 1979
see column 2, line 40 - column 5,
line 51; column 6, lines 3-12, 17-26,
31-40; column 37, line 33 - column 38,
line 32
& US, A, 4066716 (cited in the application) | 1-12

A : II Farmaco - Ed. Scientifica, volume 38,
fasc. 4, April 1983, (Pavia, IT),
P. Giori et al.: "Synthesis and
antifungal activity of pyrazole
derivatives containing sulfurated
functions", pages 274-282
see the whole document | 1-12

A : AU, A, 508225 (COMMONWEALTH SCIENTIFIC
AND INDUSTRIAL RESEARCH ORGANISATION)
13 March 1980, see claims | 1-12

* Source categories of cited documents: ¹⁰¹"A" document defining the general state of the art which is not
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other means⁵"E" document published prior to the international filing date but
later than the priority date claimed⁶"F" later document published after the international filing date
or priority date and not in conflict with the application but
said to undermine the principle or theory underlying the
invention⁷"G" document of particular relevance: the claimed invention
cannot be considered novel or cannot be considered to
involve an inventive step⁸"H" document of particular relevance: the claimed invention
cannot be considered to involve an inventive step when the
document is combined with one or more other such documents.
Such combination being obvious to a person skilled
in the art.⁹"I" document member of the same patent family

IV. CERTIFICATION

Date of the Actual Completion of the International Search

13th April 1987

Date of Mailing of the International Search Report

International Searching Authority

EUROPEAN PATENT OFFICE

Signature of Authorized Officer

III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)

Category	Citation of Document, with indication, where appropriate, of the relevant passages	Reference to Claim No.
A	US, A, 2998419 (C.L. DICKINSON et al.) 29 August 1961, see example --8; column 11, lines 14-20; claims 19,20	1-12
A	US, A, 3423424 (D.L. TREPENTER) 21 January 1969 see the whole document	1
A	Chemical Abstracts, volume 91, no. 25, 17 December 1979, (Columbus, Ohio, US), see page 181, abstract 205631p, & JP, A, 7984032 (ISHIHARA SANGYO KAISEI, LTD.) 4 July 1979	1,12

This Annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EPO file on 27/04/87

The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent document cited in search report	Publication date	Patent family member(s)
EP-A- 0201852	20/11/86	DE-A- 3602728 JP-A- 61268671 AU-A- 5736886
EP-A- 0200872	12/11/86	DE-A- 3509567- JP-A- 61215371
US-A- 3760084	18/09/73	None
US-A- 4145554	20/03/79	US-A- 4066776 NL-A- 7203453 FR-A,B 2130276 DE-A- 2212080 GB-A- 1358227 AT-B- 323156 CA-A- 998055 CH-A- 589631 CH-A- 590239 BE-A- 780675 OA-A- 4074 AU-A- 4002072 AU-B- 468250
AU-A- 508225	13/03/80	None
US-A- 2998419		None
US-A- 3423424	21/01/69	None